PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶:

(11) International Publication Number:

WO 95/32741

A61K 49/00, 51/04

A1

(43) International Publication Date:

7 December 1995 (07.12.95)

(21) International Application Number:

PCT/EP95/01958

(22) International Filing Date:

23 May 1995 (23.05.95)

(30) Priority Data:

MI94A001074

26 May 1994 (26.05.94)

IT

(71) Applicant (for all designated States except AU CA GB IE US):
BRACCO S.P.A. [IT/IT]; Via E. Folli, 50, I-20134 Milano (IT).

(71) Applicant (for AU CA GB IE only): DIBRA S.P.A. [IT/IT]; Piazza Velasca, 5, I-20122 Milano (IT).

(72) Inventors: and

- (75) Inventors/Applicants (for US only): ANELLI, Pier, Lucio [IT/IT]; Via E. Folli, 50, I-20134 Milano (IT). DE HAËN, Christoph [CH/IT]; Via E. Folli, 50, I-20134 Milano (IT). LATTUADA, Luciano [IT/IT]; Via E. Folli, 50, I-20134 Milano (IT). MOROSINI, Pierfrancesco [IT/IT]; Via E. Folli, 50, I-20134 Milano (IT). UGGERI, Fulvio [IT/IT]; Via E. Folli, 50, I-20134 Milano (IT).
- (74) Agents: MINOJA, Fabrizio; Studio Consulenza Brevettuale, Via Rossini, 8, I-20122 Milano (IT) et al.

(81) Designated States: AM, AT, AU, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IS, JP, KG, KP, KR, KZ, LK, LR, LT, LU, LV, MD, MG, MN, MX, NO, NZ, PL, PT, RO, RU, SE, SG, SI, SK, TJ, TM, TT, UA, US, UZ, VN, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG), ARIPO patent (KE, MW, SD, SZ, UG).

Published

With international search report.

Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.

(54) Title: BILE ACID CONJUGATES, DERIVATIVES THEREOF WITH METAL COMPLEXES AND RELATED USES

(57) Abstract

The invention relates to novel paramagnetic metal ion chelates and their use as contrast agents in the diagnostic technique known as "magnetic resonance imaging" (M.R.I.).

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GB	United Kingdom	MR	Mauritania
AU	Australia	GE	Georgia	MW	Malawi
BB	Barbados	GN	Guinea	NE	Niger
BE	Belgium	GR	Greece	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IE	Ireland	NZ	New Zealand
BJ	Benin	IT	Italy	PL	Poland
_		JР	-	PT	Portugal
BR	Brazil	-	Japan	RO	Romania
BY	Belarus	KE	Kenya		
CA	Canada	KG	Kyrgystan	RU	Russian Federation
CF	Central African Republic	KP	Democratic People's Republic	SD	Sudan
CG	Congo		of Korea	SE	Sweden
CH	Switzerland	KR	Republic of Korea	SI	Slovenia
CI	Côte d'Ivoire	KZ	Kazakhstan	SK	Slovakia
CM	Cameroon	LI	Liechtenstein	SN	Senegal
CN	China	LK	Sri Lanka	TD	Chad
CS	Czechoslovakia	LU	Luxembourg	TG	Togo
CZ	Czech Republic	LV	Latvia	TJ	Tajikistan
DE	Germany	MC	Monaco	TT	Trinidad and Tobago
DK	Denmark	MD	Republic of Moldova	UA	Ukraine
ES	Spain	MG	Madagascar	US	United States of America
FI	Finland	ML	Mali	UZ	Uzbekistan
FR	France	MN	Mongolia	VN	Viet Nam
GA	Gahon				•

BILE ACID CONJUGATES, DERIVATIVES THEREOF WITH METAL COMPLEXES AND RELATED USES

The present invention relates to novel paramagnetic metal ion chelates and their use as contrast agents in the diagnostic technique known as "magnetic resonance imaging" (M.R.I.). In particular, the present invention relates to bile acid conjugates with molecules endowed with a chelating capacity, as well as their complex chelates with paramagnetic metal ions and/or their salts and the use of these complexes as contrast agents for M.R.I.

5

Complexes formed of chelating agents and suitable specific metals are already used as contrastographic agents in the following diagnostic techniques: X ray imaging, nuclear magnetic resonance imaging (M.R.I.) and scintigraphy.

15 In particular, medical diagnosis using "magnetic resonance imaging" (M.R.I.), recognized as a powerful diagnostic agent in clinical practice (Stark, D.D., Bradley, W.G., Jr., Eds. "Magnetic Resonance Imaging" The C.V. Mosby Company, St. Louis, Missouri (USA), 20 1988), employs, above all, paramagnetic pharmaceutical compositions, preferably containing complex chelates of bi-trivalent paramagnetic metal ions with aminopolycarboxylic acids and/or their derivatives or analogues.

Some of them are at present in clinical use as contrast agents for M.R.I. (Gd-DTPA, N-methylglucamine salt of the gadolinium complex with diethylentriamino-pentacetic acid, MAGNEVISTR, Schering; Gd-DOTA, N-

5

10

15

20

25

30

2

methylglucamine salt of the gadolinium/1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraacetic acid complex, DOTAREM R Guerbet).

In order to illustrate the state of the art in this field, here follows a list, incomplete, though indicative, of significant patent documents: EP 71564 (Schering), US 4639365 (Sherry), US-A-4615879 (Runge), DE-A-3401052 (Schering), EP 130934 (Schering), EP 65728 (Nycomed), EP 230893 (Bracco), US-A-4826673 (Mallinckrodt), US-A-4639365 (Sherry), EP 299795 (Nycomed), EP 258616 (Salutar), WO 8905802 (Bracco).

The contrast agents listed above and on the market are designed for a wholly general use. In fact, after administration the MRI contrast agent is distributed in the extracellular spaces in different parts of the body prior to being excreted. In this sense they behave in a similar manner to iodine compounds used in X ray medical diagnosis.

Today, more than ever, the medical profession is in need of contrast agents that are aimed at specific organs, a need which is not adequately met by the products on the market at present. Especially, there is a need for contrast agents for the liver, an organ which is particularly prone to tumoral metastasis and which are almost always carcinomatose metastasis. Agents of this type should be able to provide the following results:

a) to clearly and selectively show the healthy tissue of the liver, thereby permitting the pin-pointing of small lesions such as metastasis (focal liver disease);

3

5

10

15

20

25

b) an indication of hepatic function, whereby a disease as widespread as cirrhosis of the liver may be clearly exposed;

c) a high resolution visualization of the bile ducts and of the gall bladder.

Primary hepatic carcinoma (HCC) is a pathology which has become increasingly and rapidly widespread in the last twenty years, both in the Western World and in Japan (Okuda K., Hepatology, 15, 948,1992). As a result of this, the need for a fast and efficient method of diagnosis for the detection of HCC emerges; to this purpose, Magnetic Resonance takes on a leading rôle, the proviso being the availability of a contrast agent which allows for the differentiation between the healthy hepatocites and those which are affected.

Today, only one product (AMI-HS of Advanced Magnetics, Reimer, P.; Weissleder, R. et al.; Radiology 177, 729, 1990, patent application WO-9001295) seems to possess the necessary prerequisites for the diagnosis of HCC. One is dealing with "ultra-small" particles of iron oxide (average diameter: 12nm) coated with arabinogalactose which have a particular affinity with the asialoglycoprotein receptors present on the surface of the hepatocites. However, the use of these particles brings about various side effects, especially with regard to the circulatory system. The identification of an ideal hepatospecific contrast agent is, therefore, still far off.

Among the M.R.I. contrast agents under development, both the compound known as Gd-BOPTA (BRACCO, EP 230893), and the Schering product Gd-EOB-

5

10

15

4

DTPA (EP-A-405704) turned out to be particularly suitable for the visualization of hepatic tissue, due to their characteristics of also being excreted via the bile tract.

The transport of both endogene and xenobiotic substances by means of the hepatocites and the biliary excretion mechanisms have been amply discussed in the literature, only a few basic concepts of which shall be recalled as follows (see, for example, Meier, P.J. in "Biliary Excretion of Drugs and Other Chemicals", Siegers, C.-P. and Watkins III J.B. Eds. Gustav Fischer Verlag, Stuttgart, 1991).

The passage of a molecule in bile from blood through the Disse space takes place in numerous stages that may be schematically summarized as follows:

- the molecule enters the hepatocite through the sinusoid membrane following a mechanism that may or may not be specific (mediated by a carrier or a receptor).
- inside the hepatocite the molecule may: 1) be carried unaltered and linked to an intracellular protein or inside a vesicle, 2) undergo a conjugation reaction with an enzyme and be excreted in the bile as a conjugate, 3) be enzymatically degraded inside the lisosomes.
- 25 the molecule leaves the hepatocite through the bile canaliculus membrane via a mechanism mediated by a carrier or through an exocytosis mechanism (if the molecule is carried inside the vesicle).
- If the aim is to synthesize a hepatotropic contrast agent which enters the hepatocites, the mechanisms that turn out to be the most interesting are

5

those which are mediated by a receptor or a carrier. Up to now, the following carriers have been identified and partially characterized on the membrane sinusoid:

- bile acid carriers
- 5 a bilirubin carrier

15

20

25

30

- a fatty acid carrier
- a carrier for organic cations

The first two types of carrier have been studied more in depth and the knowledge with regard to them is far more advanced.

The HCC cellular lines studied to date turn out to be made up of hepatocites which possess the bilirubin carriers. As both Gd-BOPTA and Gd-EOB-DTPA seem to penetrate the interior of the hepatocites taking advantage of said carrier, both products may not be of any help in the diagnosis of HCC, insomuch that they are not capable of differentiating between healthy and affected hepatocites.

It has been shown that in some human HCC lines the hepatocites are free from tauroalcoholic acid carriers (von Dippe, P; Levy, D.; J. Biol. Chem. 265, 5942, 1990 and cited references). It appears, therefore, that research for a contrast agent that utilizes this carrier for penetrating hepatocites is of great interest.

Patent Application (EP-A-279307, Abbott) claims polyaminocarboxylic chelant conjugates, able to complex metal ions, with different substrates, among which are the bile acids. The only illustrative complex in the case of the Patent Application is a ¹¹¹In complex of a conjugate in which a functionalized derivative of EDTA

6

is covalently linked, through an amide link, to the carboxylic function of cholic acid. The possibility of chelating paramagnetic metal ions for the use in MRI is not referred to in any way.

Another Patent Application (EP-A-417725, Hoechst) generally claims products in which a bile acid is conjugated with pharmacologically active residues such as peptids, antibiotics, antivirals, renin inhibitors medicaments for the treatment of diabetes. and the use of bile acid Recently, the results of conjugates with chlorambucil, an antitumoral agent with cytotoxic action (Kramer, W.et al.; J. Biol.Chem, 267,18598, 1992).

The present invention relates to novel compounds resulting from the conjugation of a bile acid with a chelating agent and capable of chelating the ions of bi-trivalent metals.

The present invention also relates to the complex chelates of said molecules with the ions of bitrivalent metals, as well as the salts of said chelates.

Said compounds turned out to be excellent MRI contrast agents, particularly for the "imaging" of the hepatobiliary system.

The present invention relates to the compounds of general formula (I):

A-L-B (I),

wherein

5

10

15

20

A is the residue of a bile acid, wherein by bile acid the group of the bile acids obtainable by bioconversion from cholesterol is meant,

7

particularly the acids: cholic, deoxycholic, chenodeoxycholic, ursodeoxycholic, lithocholic, and the derivatives thereof, including those with taurine and glycine;

5 L is a linker between one of the C-3, C-7, C-12 or C-24 positions of the residue of the bile acid and B, corresponding to a group of formula (II)

Y m (II)

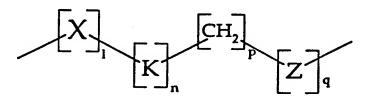
10

in which

m is an integer varying from 1 to 10, wherein for values above 1,

Y can have different meanings,

15 Y corresponds to the following succession of groups,



20 n, l and q can be 0 or 1,

p can vary from 0 to 10,

X is an O atom, a S atom, or a -NR group,

in which

R is a H atom, or a (C_1-C_5) alkyl group,

25 K is benzene ring, substituted or not, or a $-CHR_1$ group,

wherein

 R_1 is an hydrogen atom, or a -COOH group, or a -SO₃H group,

30 Z is an O atom or a S atom, or one of the -CO- or -CS- groups,

8

5

10

15

20

25

30

B is the residue of a chelating agent of bitrivalent metal ions having an atomic number varying from 20 to 31, 39, 42, 43, 44, 49, or from 57 to 83, wherein said residue can in its turn be conjugated or not, by a second chain L of formula (II), to another residue A as defined above,

with the proviso that at least one from 1, n, q, p is different from 0 and, when X and Z are both 0 or S atoms, q or n is equal to 1.

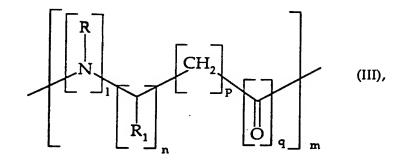
An object of the invention also are the complex chelates of said compounds of formula (I) with the bitrivalent ions of metal elements having an atomic number varying from 20 to 31, 39, 42, 43, 44, 49, or from 57 to 83, as well as the salts thereof with physiologically compatible organic bases selected from primary, secondary, tertiary amines or basic amino acids, or with inorganic bases the cations of which are sodium, potassium, magnesium, calcium or mixtures thereof, or with anions of physiologically acceptable organic acids, for example selected from acetate, succinate, citrate, fumarate, maleate, oxalate, or with anions of inorganic acids such as the ions of the halohydric acids, i.e. chlorides, bromides, iodides.

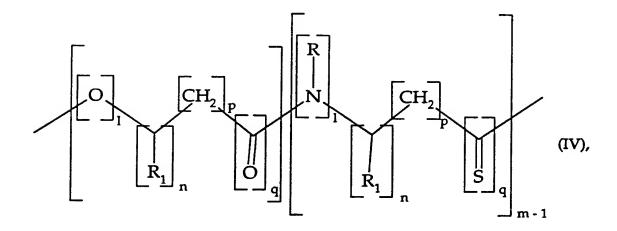
The compounds of the present invention can optionally be conjugated chemically to suitable macromolecules or inglobated into suitable carriers.

Object of the invention are also the preparation of the products of general formula (I) and of the complex salts thereof, the uses thereof and the related pharmaceutical compositions for diagnostic use.

Particularly preferred compounds of the present

invention are those in which the spacing chains L have the following general formulae (III), (IV), (V), (VI)





$$\begin{bmatrix} R \\ N \end{bmatrix}_1 \begin{bmatrix} CH_2 \\ R \end{bmatrix}_n \begin{bmatrix} CH_2 \\ N \end{bmatrix}_1 \begin{bmatrix} CH_2 \\ R \end{bmatrix}_n (V),$$

$$\begin{bmatrix} R \\ N \\ I \end{bmatrix}_{1} \begin{bmatrix} CH_{2} \\ P \end{bmatrix}_{m}$$
 (VI),

Moreover, particularly preferred are moreover the structures in which A is a residue deriving from the following bile acids or form their derivatives with taurine and glycine:

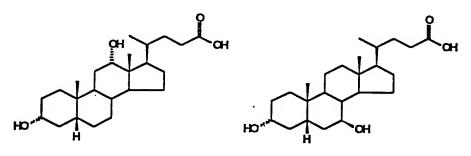
Bile acids

15

5

Cholic acid

Chenodeoxycholic acid



Deoxycholic acid

Ursodeoxycholic acid

Lithocholic acid

5

10

15

20

25

30

The bond between A and L is obtained making use either of the acidic function at the 24- position, or by functionalizing the hydroxy groups at the 3-, 7-, 12- positions, independently from the stereochemistry of the final products.

of а preferably the residue is polyaminopolycarboxylic acidic linker and derivatives thereof, particularly diethylenetriamino pentaacetic acid (DTPA), 1,4,7,10-tetraazacyclododecane-1,4,7,10tetraacetic acid (DOTA), 1,4,7,10-tetraazacyclododecane-1,4,7-triacetic acid (DO3A), [10-(2-hydroxypropyl)-1,4,7,10-tetraazacyclododecane-1,4,7-triacetic acid 4-carboxy-5,8,11-tris(carboxymethyl)-1-phe-(HPDO3A), nyl-2-oxa-5,8,11-triazatridecan-13-oic acid (BOPTA), N-[2-[bis(carboxymethyl)amino]-3-(4-ethoxyphenyl)propyl]-N-[2-[bis(carboxymethyl)amino]ethylglycine (EOB-DTPA), N, N-bis[2-[(carboxymethyl)[(methylcarbamoyl)methyl]amino]ethyl]glycine (DTPA-BMA), 2-methyl-1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraacetic $(\alpha,\alpha',\alpha'',\alpha''')$ -tetramethyl-1,4,7,10-tetraazacyclododecane-1,4,7,10-tetracetic acid (DOTMA); or B is the residue of a polyaminophosphate acidic linker or of the derivatives thereof, particularly N,N'-bis-(pyridoxal-(DPDP) 5-phosphate)ethylendiamino-N,N'-diacetic acid acid and ethylenedinitrilotetrakis(methylphosphonic)

25

30

(EDTP); or B is the residue of a polyaminophosphonic acid linker and the derivatives thereof, or polyaminophosphinic acid and the derivatives thereof, particularly 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetrakis[methylen(methylphosphonic)] acid and 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetrakis[methylen(me-thylphosphinic)] acid; or B is the residue of macrocyclic chelating agents such as texaphyrins, porphyrins and phthalocyanines.

The link to the spacing chain can be obtained by means of the acidic groups of the linker or by a suitable reactive group present in the starting linker, for example an amino group, or a functional group present on a phenyl, etc.

15 Particularly preferred reactive groups are selected from the group consisting of $-NH_2$, -NCS, $-NHCSNHNH_2$, $-NHCSNH(CH_2)_2NH_2$, -NCO, $-NHNH_2$, $-NHCONHNH_2$, -CHO.

Particularly preferred are the structures in which

A is a residue of cholic acid, B is a residue of the

linker BOPTA, of DTPA or of DOTA.

Metal ions suitable to form complex salts with the chelating agents of general formula (I) are mainly the bivalent or trivalent ions of the elements having atomic numbers varying from 20 to 31, 39, 42, 43, 44, 49, or from 57 to 83; particularly preferred are $Fe^{(2+)}$, $Fe^{(3+)}$, $Cu^{(2+)}$, $Cr^{(3+)}$, $Gd^{(3+)}$, $Eu^{(3+)}$, $Dy^{(3+)}$, $La^{(3+)}$, $Yb^{(3+)}$ or $Mn^{(2+)}$ or also radioisotops such as 51_{Cr} , 67_{Ga} , 68_{Ga} , 111_{In} , 99_{Tc} , 140_{La} , 175_{Yb} , 153_{Sm} , 166_{Ho} , 90_{Y} , 149_{Pm} , 177_{Lu} , 47_{Sc} , 142_{Pr} , 159_{Gd} , 212_{Bi} .

The compounds of general formula (I) can be

13

prepared with synthesis methods conventionally known in industrial technology. Particularly, MRI contrast agents conjugated with bile acids can be prepared by means of a convergent synthesis which comprises:

- 5 1) synthesis of a functionalized ligand i.e. of a ligand capable of coordinating one paramagnetic metal ion and at the same time of bind stably to the bile acid by means of a suitable functional group;
- 10 2) synthesis of a functionalized bile acid;
 - 3) coupling reaction between two different syntons;
 - 4) cleavage of any protective groups;
 - 5) complexation of the paramagnetic metal ion.

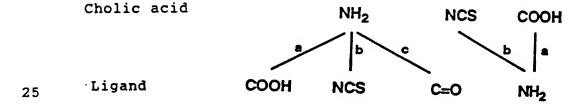
In the following Scheme 1, some of the functional 15 groups most easily obtainable respectively on cholic on acid and the ligand, as well as the mutual possibilities to react to give stable bonds are · reported by way of example.

Scheme 1

20

30

Functional groups



As it can be observed, the conjugation of the two syntons is carried out through three different known binding methods, widely used in synthesis (see Brinkley, M., Bioconjugate Chem. 1992, 3, 2), which involve formation of an amide (path a), of a thiourea

(path b) or, through reduction of the intermediate imine, of an amine (path c).

The functional groups of the two syntons of Scheme 1 are moreover liable to further modifications before the binding reaction, for example by reaction with suitable bifunctional spacers.

An example of ligands corresponding to point 1) described above is represented by the molecules in Scheme 2.

10 Scheme 2

5

10

15

By way of non-limiting example, the synthesis of ligands C and D, the latter being the glycine derivative (Scheme 3), can be cited herein.

The synthesis of t-butyl 2-bromo-3-[(4-nitrophenyl)methoxy]propionate was performed according to the procedure described by P. L. Rings et al., Synth. Commun., 23, 2639, 1993. In a similar way, starting from t-butyl 2-bromo-4-[(4-nitrophenyl)]butanoate (prepared according to the procedure described in Kruper W.J.; Rudolf P.R.; Langhoff C.A. J. Org. Chem., 58, 3869, 1993), a ligand which involves no benzyloxy groups can be prepared.

In Scheme 3 the t-butyl esters are shown, but they can easily be substituted by other alkyl groups.

- The synthesis continues with the condensation of the a-bromopropionic intermediate with diethylenetriamine and the subsequent carboxymethylation with t-butyl a-bromoacetate under the usual conditions known in literature.
- The reduction of the nitro group is performed by means of hydrogen using 10% Pd/C as the catalyst. At this point, the key synton is available for binding through the amino group present on an aromatic ring of the ligand with the acidic groups of the bile acids or with derivatives thereof in which carboxylic groups are present.

16

An example of functionalization of the steroid can be represented by the synthesis of cholic acid 3β -amino derivative according to Scheme 4, using in an original way the Mitsunobu reaction (Review, Synthesis, 1, 1981) which allows the selective transformation of the 3d hydroxyl group into the corresponding 3β azido group.

30

Scheme 4

COOMe

ETOOC-N=N-COOEt

Ph₃P/(PhO)₂PON₃

Ph3P/THF/H2O

.COOMe 10

COOMe

15 Intermediate G can be used as such or easily be changed into other intermediates as much interesting, as evidenced in Scheme 5.

Scheme 5

20

5

25

18

5

10

15

20

25

An example of binding reaction between the components A and B of general formula (I) of the compounds of the present invention, is the formation of the amido bond between the steroid carboxylic group at the 24- position and the amino group present in the ligand, for example C and D of Scheme 2.

The reaction is preferably activated by the addition of diethoxyphosphoryl cyanide (DEPC), according to the procedure described for the peptide synthesis (Shioiri,T et al., Tetrahedron, 32, 2211, 1976). The reaction with DEPC takes place preferably in a dipolar aprotic solvent, such as dimethylformamide (DMF) or dimethylacetamide (DMA) or in a mixture thereof, at a temperature varying from -5°C to 40°C, preferably from 0°C to 25°C.

A further example of binding reaction makes use of the formation of a Schiff base between the ligands of type E and F and a suitable steroid derivative, for example the derivative J obtained from 3β -aminocholic derivative according to Scheme 5.

The aldehyde group of the ligand reacts with the amino group present on the steroid and subsequently the amino derivative is reduced with NaBH₃CN, according to a well-known procedure of the literature (C.F. Lane, Synthesis, 135, 1975).

The choice of diversifying the ester groups present in both components of the binding reaction, allows for the modulation of the hydrolysis thereof in different synthesis steps.

30 The conversion of the ester groups of t-butyl type into acidic groups takes place in acid solution. The

19

resulting solution is adjusted to controlled pH thus allowing the simultaneous formation of the desired complex by addition of the stoichiometric amount of metal, in the form of oxide or salt.

The hydrolysis reaction of the ester groups of methyl type takes place preferably in the presence of a suitable organic or inorganic base such as sodium hydroxide, potassium hydroxide, potassium carbonate or, for example, lithium hydroxide at a pH value varying from 8 to 12, preferably between 0°C and 100°C, more preferably between 0°C and 50°C.

The possible conjugation with the amino acids taurine and glycine takes place according to the procedure described in Tserng, K-Y.; Hachey, D.L.; Klein, P.D. J. Lipid Res. 1977, 18, 404.

15

25

Finally, the formation of the metal complex salt is preferably carried out in water or in a suitable water-alcohol mixture, whereas the temperature can vary from 25°C to 100°C, preferably from 40°C to 80°C.

The choice of the metal ion and of any neutralizing ions is strictly related to the use of the complex to be prepared.

The novel compounds of the present invention proved to have a good tolerability; moreover their water solubility and the low osmolality of the solutions are another important feature making them particularly suited for the use in nuclear magnetic resonance.

The <u>in vitro</u> relaxivity data evidenced for the compounds of the present invention turned out to be quite good. By way of non-limiting example, the r_1 and

25

30

r₂ values found for two of the preferred compounds of the invention, i.e. the 4-carboxy-5,8,11-tris(carboxymethyl) $-1-[4-[[[(3a,5\beta,7a,12a)-3,7,12-trihydroxy-24$ oxocholan-24-yl]amino]acetyl]amino]phenyl]-2-oxa-5,8,11-triazatridecan-13-oic acid gadolinium complex 5 salified with 1-deoxy-1-(methylamino)-D-glucitol (1:2), $[[10-[2-0xo-2-[3-[2-[(3a,5\beta,7a,12a)-3,7,12-tri$ hydroxy-24-oxocholan-24-yl]amino]ethyl]amino]propyl]amino]ethyl]-1,4,7,10-tetraazacyclododecan-1,4,7-triacetoate(3-)]gadolinate(0)] hydrogen compound with HCl 10 (1:1), are reported in EXAMPLE 19, compared with the data available for paramagnetic compounds marketed under the trade marks MAGNEVIST^R (Schering) DOTAREMR (Guerbet), or with the data related to Gd-BOPTA and to the Gd^{3+} ion as such. 15

Both soluble and less soluble compounds are suited for the oral or enteral administrations and, therefore, particularly for the gastrointestinal tract imaging.

For the parenteral administration, they are preferably formulated as sterile aqueous solutions or suspensions, whose pH can range, for example, from 6.0 to 8.5.

Said aqueous solutions or suspensions can be administered in concentrations varying from 0.002 to 1.0 Mol.

Said formulations can be freeze-dried and provided as such for the extemporary use. For the gastrointestinal use or for the injection in body cavities, said agents can be formulated as solutions or suspensions containing appropriate additives suitable, for example, to control viscosity.

21

5

15

For the oral administration, they can be formulated according to preparation methods conventionally used in pharmaceutical technique, possibly also as coated formulations to obtain an additional protection against the stomach acidic pH, thus preventing the chelated metal ion from release, which takes place particularly at the pH values typical of gastric juices.

Other excipients, such as sweeteners and/or flavouring agents, can also be added, according to known techniques of pharmaceutical formulations.

As far as the diagnostical use of the chelates of the present invention is concerned, they can also be used as both contrast media and therapeutical agents, in nuclear medicine.

In this case, however, the metal ion which is chelated is a radioisotope, for example ^{51}Cr , ^{67}Ga , ^{68}Ga , ^{111}In , ^{99m}Tc , ^{140}La , ^{175}Yb , ^{153}Sm , ^{166}Ho , ^{90}Y , ^{149}Pm , ^{177}Lu , ^{47}Sc , ^{142}Pr , ^{159}Gd and ^{212}Bi .

Preferred inorganic base cations possibly suitable to salify the complex chelates of the present invention comprise particularly the alkali or alkaline-earth metal ions such as potassium, sodium, calcium, magnesium, and mixtures thereof.

Preferred organic base cations suitable for the above mentioned purpose comprise, <u>inter alia</u>, those of primary, secondary and tertiary amine, such as ethanolamine, diethanolamine, morpholine, glucamine, N-methylglucamine, N,N-dimethylglucamine.

Preferred inorganic acid anions possibly suitable to salify the complex chelates of the present invention

10

15

20

comprise, particularly, the halohydric acid ions, such as chlorides, bromides, iodides or other ions such as sulfate.

Preferred organic acid anions for the above mentioned purpose comprise those of acids conventionally used in pharmaceutical techniques for the salification of alkali substances, such as acetate, succinate, citrate, fumarate, maleate.

Preferred amino acid cations and anions comprise, for example, those of taurine, glycine, lysine, arginine or ornithine or of the aspartic and glutamic acids.

The complex chelates conjugated with the bile acids, object of the present invention, can also be inglobated into liposomes or be components of their chemical structure and be used as mono- or multi-lamellar vescicles.

A non-limiting list of preferred compounds of the invention (described in the experimental part) is reported in the following to illustrate further the present invention.

COMPOUND 1 (EXAMPLE 1)

30 [[4-carboxy-5,8,11-tris(carboxymethyl)-1-[4-[[[(3α,5β,7α,12α)-3,7,12-trihydroxy-24-oxocholan-24-

23

amino]acetyl]amino]phenyl]-2-oxa-5,8,11-triazatridecan13-oic acid

COMPOUND 2 (EXAMPLE 2)

5

10

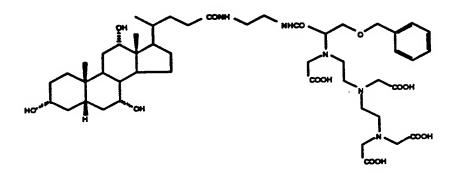
[[4-carboxy-5,8,11-tris(carboxymethyl)-1-[4-[[(3α , 5β , 7α ,12 α)-3,7,12-trihydroxy-24-oxocholan-24-yl]amino]phenyl]-2-oxa-5,8,11-triazatridecan-13-oic acid

15

COMPOUND 3 (EXAMPLE 3)

20

25



[[3,6,9-tris(carboxymethyl)-10-(phenylmethoxy)methyl-11-oxo-14-[[3 α ,5 β ,7 α ,12 α)-3,7,12-trihydroxy-24-oxocholan-24-yl]amino]-3,6,9,12-tetraazatetradecanoic acid

COMPOUND 4 (EXAMPLE 4)

[[10-[2-0x0-2-[[3-[[2-[[(3a,5\beta,7a,12a)-3,7,12-trihy-droxy-24-oxocholan-24-yl]amino]ethyl]amino]propyl]-amino]ethyl]-1,4,7,10-tetraazacyclododecan-1,4,7-triacetic acid

COMPOUND 5 (EXAMPLE 5)

 $\begin{tabular}{ll} [[(3\beta,5\beta,7\alpha,12\alpha)-3-[[13-carboxy-6,9,12-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-trazatridecyl] amino]-7,12-dihydroxy-cholan-24-oic acid \\ \begin{tabular}{ll} [[(3\beta,5\beta,7\alpha,12\alpha)-3-[[13-carboxy-6,9,12-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-tris(carboxyme-thyl)-3,6,9,12-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-tris(carboxyme-thyl)-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-dioxo-1,4-d$

COMPOUND 6 (EXAMPLE 6)

[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[17-carboxy-10,13,16-tris(carboxy-methyl)-8-oxo-9-[(phenylmethoxy)methyl]-3,7,10,13,16-pentaazaheptadecyl]oxy]-7,12-dihydroxy-cholan-24-oic acid

COMPOUND 7 (EXAMPLE 7)

5

15

20

25

10

10

15

20

25

30

25

[[(3\beta,5\beta,7\alpha,12\alpha)-7,12-dihydroxy-3-[2-[[[[4-[4,12-bis(carboxy)-5,8,11-tris(carboxymethyl)-2-oxa-5,8,11-triazadodecyl]phenyl]amino]thioxomethyl]amino]ethoxy]-cholan-24-oic acid

COMPOUND 8 (EXAMPLE 4)

(3β,5β,7α,12α)-7,12-dihydroxy-3-[[[[3-[[[4,7,10-tris-(carboxymethyl)-1,4,7,10-tetraazacyclodec-1-yl]acetyl]-amino]propyl]amino]acetyl]amino]-cholan-24-oic acid

COMPOUND 9 (EXAMPLE 8)

[[3,6,9-tris(carboxymethyl)-10-[(phenylmethoxy)methyl]11-oxo-17-[[(3α,5β,7α,12α)-3,7,12-trihydroxy-24-oxocholan-24-yl]amino]-3,6,9,12-tetraazaoctadecanedioic acid

COMPOUND 10 (EXAMPLE 9)

[[(3 β ,5 β ,7 α ,12 α)-3-[[13-carboxy-6,9,12-tris(carboxyme-thyl)-1,4-dioxo-3,6,9,12-tetraazatridecyl]amino]-7,12-dihydroxy-cholan-24-oic acid

COMPOUND 11 (EXAMPLE 10)

5

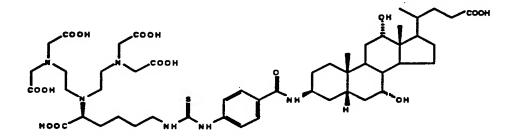
10

[$(3\beta,5\beta,7\alpha,12\alpha)-(3'\beta,5'\beta,7'\alpha,12'\alpha)-3,3'-[[6,9,12-tris-(carboxymethyl)-1,4,14,17-tetraoxo-3,6,9,12,15-penta-azaheptadecan-1,17-diyl]bisimino]bis[7,12-dihydroxycho-lan-24-oic acid]$

COMPOUND 12 (EXAMPLE 11)

20

15



25

[[[3\beta(S),5\beta,7\alpha,12\alpha]-7,12-dihydroxy-3-[[4-[[[5-[bis[2-[bis(carboxymethyl)amino]ethyl]amino]-5-carboxypentyl]-amino]thioxomethyl]amino]benzoyl]amino]-cholan-24-oic acid

27
COMPOUND 13 (EXAMPLE 12)

[[[3β(S),5β,7α,12α]-7,12-dihydroxy-3-[[4-[[5-[bis[210 [bis(carboxymethyl)amino]ethyl]amino]-5-carboxypentyl]amino]-1,4-dioxobutyl]amino]-cholan-24-oic acid
COMPOUND 14 (EXAMPLE 13)

[[(3β,5β,7α,12α)-7,12-dihydroxy-3-[[4-[[2-[[4-[4,12-bis(carboxy)-5,8,11-tris(carboxymethyl)-2-oxa-5,8,11-triazadodecyl]phenyl]amino]-2-oxoethyl]amino]-1,4-dio-xobutyl]amino]-cholan-24-oic acid;

COMPOUND 15 (EXAMPLE 5)

10

28

3,6,9-tris(carboxymethyl)-14-[[(3\beta,5\beta,7\alpha,12\alpha)-7,12-dihydroxy-24-oxo-24-[(2-sulfoethyl)amino]-cholan-3-yl]amino]-11,14-dioxo-10-(phenylmethoxy)methyl-3,6,-9,12-tetraazatetradecanoic acid

COMPOUND 16 (EXAMPLE 5)

N-[(3β,5β,7α,12α)-3-[[13-carboxy-6,9,12-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-trazatridecyl]amino]-7,12-dihydroxy-24-oxocholan-24-yl]glycine

COMPOUND 17 (EXAMPLE 5)

25 (3β,5β,7α)-3-[[13-carboxy-6,9,12-tris(carboxymethyl)1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-tetraazatridecyl]amino]-7-hydroxy-cholan-24-oic acid

29

5

15

COMPOUND 18 (EXAMPLE 5)

(3β,5β,12α)-3-[[13-carboxy-6,9,12-tris(carboxymethyl)1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-tetraazatridecyl]amino]-12-hydroxy-cholan-24-oic acid

COMPOUND 19 (EXAMPLE 5)

20 (3β,5β)-3-[[13-carboxy-6,9,12-tris(carboxymethyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-tetraazatride-cyl]amino]-cholan-24-oic acid

COMPOUND 20 (EXAMPLE 5)

30

 $(3\beta,5\beta,7\alpha,12\alpha)-3-[[17-carboxy-10,13,16-tris(carboxyme-thyl)-1,8-dioxo-9-[(phenylmethoxy)methyl]-7,10,13,16-tetraazaheptadecyl]amino]-7,12-dihydroxy-cholan-24-oicacid$

COMPOUND 21 (EXAMPLE 4)

 $(3\beta,5\beta,7\alpha,12\alpha)-7$,12-dihydroxy-3-[[3-[[4,7,10-tris(carboxymethyl)-1,4,7,10-tetraazacyclododec-1-yl]acetyl]-amino]propyl]amino]-cholan-24-oic acid

COMPOUND 22 (EXAMPLE 9)

3,6,9-tris(carboxymethyl)-14-[[(3 β ,5 β ,7 α ,12 α)-7,12-dihydroxy-24-oxo-24-[(2-sulfoethyl)amino]-cholan-3-yl]amino]-11,14-dioxo-3,6,9,12-tetraazatetradecanoic acid

30

5

10

31

COMPOUND 23 (EXAMPLE 9)

10 $[(3\beta,5\beta,7\alpha,12\alpha)-3-[[17-carboxy-10,13,16-tris(carboxyme-thyl)-1,8-dioxo-7,10,13,16-tetraazaheptadecyl]amino]-7,12-dihydroxy-cholan-24-oic acid$

COMPOUND 24 (EXAMPLE 9)

(17S)-3,6,9-tris(carboxymethyl)-11-oxo-17-[[(3 β ,5 β ,-7 α ,12 α)-3,7,12-trihydroxy-24-oxocholan-24-yl]amino]-3,6,9,12-tetraazaoctadecanedioic acid

COMPOUND 25 (EXAMPLE 14)

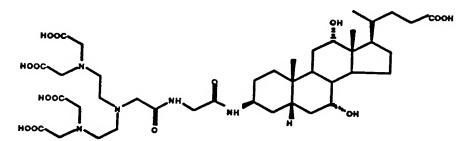
PCT/EP95/01958 WO 95/32741

32

 $[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[6-[[[bis[2-[bis(carboxymethyl)ami$ no]ethyl]amino]acetyl]amino]-1-oxohexyl]amino]-7,12-dihydroxy-cholan-24-oic acid

COMPOUND 26 (EXAMPLE 14)

5

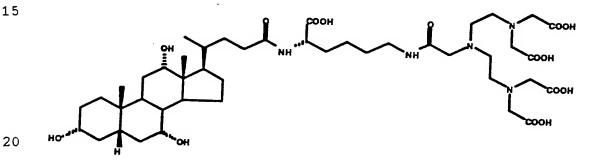


10

 $(3\beta,5\beta,7\alpha,12\alpha)-3-[[[[[bis[2-[bis(carboxymethyl)amino]$ ethyl]amino]acetyl]amino]acetyl]amino]-7,12-dihydroxycholan-24-oic acid

COMPOUND 27 (EXAMPLE 14)

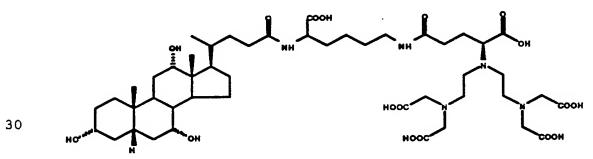
15



N⁶-[[bis[2-[bis(carboxymethyl)amino]ethyl]amino]ace $tyl]-N^2-[(3\alpha,5\beta,7\alpha,12\alpha)-3,7,12-trihydroxy-24-oxocholan-$ 24-yl]-L-lysine

25

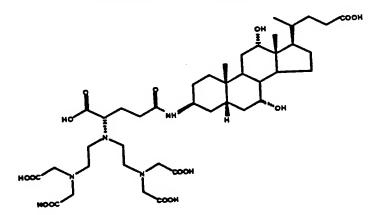
COMPOUND 28 (EXAMPLE 15)



33

COMPOUND 29 (EXAMPLE 15)

5



10

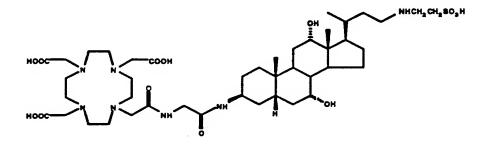
15

 $[3\beta(S),5\beta,7\alpha,12\alpha]-3-[4-carboxy-4-[bis[2-[bis(carboxyme-thyl)amino]-1-oxobutyl]amino]-7,12-dihydro-xy-cholan-24-oic acid$

COMPOUND 30 (EXAMPLE 16)

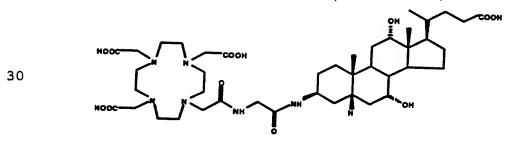
20

25



[[10-[2-[[2-[[(3\beta,5\beta,7\alpha,12\alpha)-7,12-dihydroxy-24-oxo-24-[(2-sulfoethyl)amino]cholan-3-yl]amino]-2-oxoethyl]amino]-2-oxoethyl]-1,4,7,10-tetraazacyclododecane-1,4,7triacetic acid

COMPOUND 31 (EXAMPLE 16)



SUBSTITUTE SHEET (RULE 26)

(3β,5β,7α,12α)-3-[[[[[4,7,10-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecyl]acetyl]amino]acetyl]amino]-7,12-dihydroxy-cholan-24-oic acid

COMPOUND 32 (EXAMPLE 16)

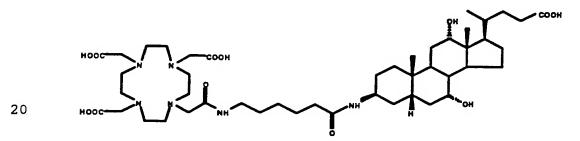
5

 N^2 -[(3a,5 β ,7a,12a)-3,7,12-trihydroxy-24-oxocholan-24-yl]- N^6 -[[4,7,10-tris(carboxymethyl)-1,4,7,10-tetraaza-cyclododecyl]acetyl]-L-lysine

15

10

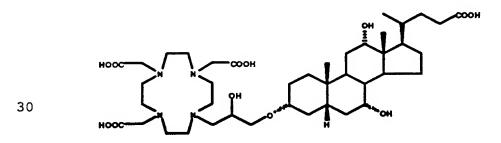
COMPOUND 33 (EXAMPLE 16)



 $(3\beta, 5\beta, 7\alpha, 12\alpha)-3-[[6-[[[4,7,10-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecyl]acetyl]amino]-1-oxohe-xyl]amino]-7,12-dihydroxy-cholan-24-oic acid$

25

COMPOUND 34 (EXAMPLE 17)



35

[[(3α,5β,7α,12α)-3-[[3-[4,7,10-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecyl]-2-hydroxypropyl]oxy]-7,12-dihydroxy-cholan-24-oic acid

COMPOUND 35 (EXAMPLE 18)

5

10

20

25

30

[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[5-[4,7,10-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecyl]-4-hydroxy-1-oxopentyl]-amino]-7,12-dihydroxy-cholan-24-oic acid

It is intended that all the matter contained in the following section shall be interpreted as illustrative and not in a limiting sense.

EXAMPLE 1

[[4-Carboxy-5,8,11-tris(carboxymethyl)-1-[4-[[[[(3α,-5β,7α,12α)-3,7,12-trihydroxy-24-oxocholan-24-yl]amino]-acetyl]amino]phenyl]-2-oxa-5,8,11-triazatridecan-13-oa-te(5-)]gadolinate(2-)] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:2)

A) N-[2-[(2-aminoethyl)amino]ethyl]-O-(4-nitrophe-nyl)methyl-D,L-serine t-butyl ester

A solution of 14 g of t-butyl 2-bromo-3-[(4-nitrophenyl)methoxy]propanoate (prepared according to the procedure described by P. L. Rings et al., Synth. Commun., 1993, 23, 2639) (0.0389 mol), in 30 ml of acetonitrile was added with a solution of 20 g of diethylenetriamine (0.19 mol) in 20 ml of acetonitrile

kept at 0-5°C and under inert atmosphere. The solution was then heated to 35°C for 4 h. The solvent was evaporated under vacuum and 100 ml of a NaCl saturated solution were added to the residue. The solution was extracted with $\rm Et_2O$; the organic phase was washed with $\rm H_2O$, dried and concentrated under vacuum to obtain 12 g of the desired product (0.031 mol).

Yield: 80%

5

TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

10 Eluent: $CHCl_3$: CH_3OH : 25% NH_4OH (w/w) = 10:2:0.5 (v/v/v)

Detector: 0.5% KMnO $_4$ in 0.1N NaOH R_f = 0.5 The $^1\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

15 B) 4-[(1,1-dimethylethoxy)carbonyl]-5,8,11-tris[2-(1,1-dimethylethoxy)-2-oxoethyl]-1-(4-nitrophe-nyl)-2-oxa-5,8,11-triazatridecan-13-oic acid 1,1-dimethylethyl ester

A solution of 11 g of compound A) (0.029 mol) and 20 31.17 g of diisopropylethylamine (0.29 mol) in 50 ml of 1,2-dichloroethane, kept at 0-5°C and under atmosphere, was added with 28.28 g of t-butyl bromoacetate (0.145 mol). The reaction mixture was kept under stirring at room temperature for 16 h. After 25 cooling to 0°C, the solution was filtered and the solvent was evaporated under reduced pressure. residue was taken up into AcOEt and H₂O. After evaporation of the solvent, the residue was purified by column chromatography, to obtain 14,8 g of the desired 30 product (0.018 mol).

Yield: 62%

37

TLC: Carrier: silica gel plates 60 F_{254} Merck

Eluent: n-hexane : EtOAc = 7:3 (v/v)

Detector: 0.5% $KMnO_4$ in 0.1N NaOH $R_f = 0.5$

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

- C) 4-[(1,1-dimethylethoxy)carbonyl]-5,8,11-tris[2 (1,1-dimethylethoxy)-2-oxoethyl]-1-(4-aminophe nyl)-2-oxa-5,8,11-triazatridecan-13-oic acid 1,1 dimethylethyl ester
- A solution of 13.7 g of compound B) (0.0163 mol) in 200 ml EtOH was added with 1.37 g of 10% palladium carbon and the mixture was hydrogenated at room temperature and normal pressure for 1 h. The suspension was filtered from the catalyst through Millipore R (0.5
- 15 μm) and the solvent was evaporated under reduced pressure. The residue was purified by flash chromatography to obtain 10.7 g of the desired product (0.0132 mol).

Yield: 81%

20 HPLC titre: 99% (% area)

Stationary phase: column E. Merck Lichrosorb Select B 5 μm ; 250 x 4 mm

Mobile phase: gradient elution

A = aqueous solution 0.01M KH_2PO_4 and 0.017M H_3PO_4

 $B = CH_3CN$

min	% A	% B
0	95	5
30	20	80
35	95	5

30 Flow rate: 1 ml min⁻¹

Temperature: 45°C

Detector (UV): 245 nm

TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

Eluent: $n-hexane : Et_2O : i-PrOH = 70:25:5 (v/v/v)$

Detector: UV lamp (254 nm) $R_f = 0.15$

- 5 The 1 H-NMR, 13 C-NMR, IR and MS spectra are consistent with the indicated structure.
 - D) 4-[(1,1-dimethylethoxy)carbonyl]-5,8,11-tris[2-(1,1-dimethylethoxy)-2-oxoethyl]-1-[4-[[[[(1,1-dimethylethoxy)carbonyl]amino]acetyl]amino]phenyl]2-oxa-5.8,11-triazatridecan-13-oic acid 1,1-dimethylethoxy
- 2-oxa-5,8,11-triazatridecan-13-oic acid 1,1-dimethylethyl ester

A solution of 4.95 g of compound C) (0.0061 mol), and 2.13 g of N-(t-butoxycarbonyl)glycine (marketed product) (0.0122 mol) and 1.35 g of triethylamine (0.0134 mol) in 50 ml of DMF, kept under stirring at 0°C, was added drop by drop with 2.18 g of diethoxyphosphoryl cyanide (0.0134 mol), under inert atmosphere, in 15 minutes. When the addition was over, the mixture was left to warm to room temperature. After 120 h the mixture was diluted with AcOEt and washed with a NaCl saturated solution. The organic phase was then washed with a 10-5N HCl solution, with H₂O and evaporated under reduced pressure. The residue was purified by flash chromatography to obtain 3.02 g of

Yield: 51%

25

HPLC titre: 98% (in % area)

the desired product (0.0031 mol).

Stationary phase: column E. Merck Lichrosorb Select B, $5 \ \mu m$; $250 \ x \ 4 \ mm$

30 Mobile phase: gradient elution

A = aqueous solution 0.01M KH_2PO_4 and 0.017M H_3PO_4

39

 $B = CH_3CN$

min	% A	% B
0	95	5
30	20	80
45	20	80

Flow rate: 1 ml min^{-1}

Temperature: 45°C

5

15

20

25

30

Detector (UV): 245 nm

TLC: Carrier: silica gel plates 60 F_{254} Merck

10 Eluent: $n-hexane : Et_2O : i-PrOH = 70:25:5 (v/v/v)$

Detector: UV lamp (254 nm) $R_f = 0.15$

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

E) 4-Carboxy-5,8,11-tris(carboxymethyl)-1-[4-[[[-(3α,5β,7α,12α)-3,7,12-trihydroxy-24-oxocholan-24yl]amino]acetyl]amino]phenyl]-2-oxa-5,8,11-triazatridecan-13-oic acid

A solution of 35.09 g of compound D) (0.036 mol) and 270 ml of anisole in 340 ml of CH2Cl2 was added drop by drop, at 0°C, with 167 ml of trifluoroacetic acid in a time of 2 h. When the addition was over the mixture was left to warm to room temperature reacting for a 3 day total time. The reaction mixture was evaporated under reduced pressure. The residue was taken up into $\mathrm{CH}_2\mathrm{Cl}_2$. The residue was then suspended in $\rm H_2O$, neutralized at 0°C with 25% $\rm NH_4OH$ (w/w) and extracted with ether ethyl. The aqueous phase was evaporated under reduced pressure to obtain a residue purified by flash chromatography. resulting solid was dissolved at room temperature in a $\rm H_2O/DMF$ mixture (5 : 8 = $\rm v/v$) and reacted with 21.85 g

40

of cholic acid N-succinimidyl ester (prepared according to the procedure described by Okahata, Y; Ando, R.; Kunitake, T., Bull. Chem. Soc. Jpn., 1979, 52, 3647-3653) (0.043 mol) added in small portions to the solution. After 30 h the reaction mixture was evaporated under reduced pressure and the residue was purified by flash chromatography. The product was dissolved in 1N HCl and eluted through an Amberlite^R XAD-16 polystyrene resin, to obtain 10.08 g of the desired product (0.010 mol).

Yield: 29% m.p.: 154-156°C (dec.)

K.F. titre: 1.79% (w/w)

HPLC titre: 98% (in % area)

Stationary phase: column E. Merck Lichrosorb Select B,

15 5 µm; 250 x 4 mm

Mobile phase: Gradient elution

A = aqueous solution 0.01M $\mathrm{KH_2PO_4}$ and 0.017M $\mathrm{H_3PO_4}$ B = $\mathrm{CH_3CN}$

	min	% A	% B
20	0	95	5
	30	20	80
	45	20	80

Flow rate: 1 ml min^{-1}

Temperature: 45 °C

25 Detector (UV): 245 nm

Elemental analysis C H N
% calc.: 59.06 7.54 7.18
% found: 58.22 7.78 7.13

TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

30 Eluent: CH_2Cl_2 : MeOH: 25% NH_4OH (w/w) = 6:3:0.7 (v/v/v)

Detector: UV lamp (254 nm) or AcOH : conc. $\rm H_2SO_4$: p-anisaldehyde $\rm R_f$ = 0.21

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

5 F) Title compound

10

15

9.04 g of compound E) (8.9 mmol) were suspended in 200 ml of $\rm H_2O$ and pH was adjusted to 6.5 with 21.9 ml of 1N meglumine (21.9 mmol) to obtain a complete dissolution. Then a solution of 3.29 g of $\rm GdCl_3$.6 $\rm H_2O$ (8.9 mmol) in 40 ml of $\rm H_2O$ was dropped therein, maintaining at pH 6.5 by addition of 1N meglumine (43.7 ml total; 43.7 mmol); when pH remained constant without addition of meglumine, the mixture was filtered through a Millipore HA filter (0.45 µm) and subjected to nanofiltration. pH of the retentate was adjusted to 7 with meglumine and the solution was concentrated to dryness. The vitreous residue was ground and dried to

Yield: 96% m.p.: 178-180°C (dec.)

obtain 13 g of the desired product (8.5 mmol).

20 K.F. titre: 6.84% (w/w)

HPLC titre: 98% (in % area)

Stationary phase: Column E. Merck Superspher RP 18; 5 μm ; 250 x 4 mm

Mobile phase: Gradient elution

A = 0.05 M KH_2PO_4 aqueous solution adjusted to pH 3.5 with H_3PO_4

 $B = CH_3CN$

	min	8 A	% B
	0	70	30
30	15	70	30
	30	50	50

42

Flow rate: 1 ml min^{-1}

Temperature: 40°C

Detector (UV): 245 nm

Elemental analysis C H Gd N Cl

5 % calc.: 48.96 6.89 10.34 6.45

% found: 45.88 7.12 9.70 6.06 < 0.1

The IR and MS spectra are consistent with the indicated structure.

EXAMPLE 2

- [[4-Carboxy-5,8,11-tris(carboxymethyl)-1-[4-[[(3α,5β,-7α,12α)-3,7,12-trihydroxy-24-oxocholan-24-yl]amino]phe-nyl]-2-oxa-5,8,11-triazatridecan-13-oate(5-)]gadolina-te(2-)] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:2)
- A solution of 8.4 g of cholic acid (marketed product) (20.6 mmol) in 20 ml of DMF at 10°C, was added with 2.25 g of triethylamine (22.2 mmol) and 13.9 g of 4-[(1,1-dimethylethoxy)carbonyl]-5,8,11-tris[2-(1,1-dimethylethoxy)-2-oxoethyl-1-(4-aminophenyl)-2-oxa-
- 5,8,11-triazatridecan-13-oic acid 1,1-dimethylethyl ester (prepared according to the procedure described in EXAMPLE 1) (17.2 mmol) dissolved in 40 ml of DMF, to obtain a kind of gel. Then 4.19 g of diethoxyphosphoryl cyanide (23.9 mmol) in 5 ml of DMF were dropped therein, at 7°C and in 10 min. When the addition was over, the solution was homogeneous again. After 2 h at

43

 7°C and 2 h at room temperature, the reaction mixture was poured into H_2O and extracted with Et_2O . The organic phase was washed with a 5°NaHCO_3 solution, then with a NaCl saturated solution, dried and evaporated under reduced pressure. The residue was purified by flash chromatography to obtain 9.5 g of the desired product (7.9 mmol).

Yield: 46%

HPLC titre: 92% (in % area)

10 Stationary phase: Column E. Merck Lichrosorb Select B; 5 μm; 250 x 4 mm

Mobile phase: Gradient elution

A = 0.01M $\mathrm{KH_2PO_4}$ and 0.017M $\mathrm{H_3PO_4}$ aqueous solution B = $\mathrm{CH_3CN}$

15	min	% A	% B
	0	95	5
	30	20	80
	4.5	20	80

Flow rate:

 1 ml min^{-1}

20 Temperature:

30 °C

Detector (UV): 245 nm

TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

Eluent: AcOEt : i-PrOH= 95 : 5 (v/v)

Detector: UV lamp (254 nm) $R_f = 0.42$

- 25 The $^{1}H-NMR$, $^{13}C-NMR$, IR and MS spectra are consistent with the indicated structure.
 - B) 4-Carboxy-5,8,11-tris(carboxymethyl)-1-[4-[[(3 α ,-5 β ,7 α ,12 α)-3,7,12-trihydroxy-24-oxocholan-24-yl]-amino]phenyl]-2-oxa-5,8,11-triazatridecan-13-oic

30 acid

6.05 g of compound A) (5 mmol) were dissolved in

44

40 ml of $\mathrm{CH_2Cl_2}$, the solution was cooled to 0°C and 20 ml of $\mathrm{CF_3COOH}$ were dropped slowly (1 h). The reaction mixture was left under stirring at room temperature for 24 h, the solvent was evaporated off under reduced pressure and the residue, taken up into $\mathrm{CH_2Cl_2}$, was evaporated again to remove completely $\mathrm{CF_3COOH}$. The resulting residue was taken up into $\mathrm{CH_2Cl_2}$ and purified by flash chromatography. The product was dissolved in 1N HCl and eluted through an Amberlite^R XAD-16 polystyrene resin, to obtain 2.9 g of the desired product (3.15 mmol).

Yield: 60%

Elemental analysis C H N

% calc.: 60.11 7.68 6.09

15 % found: 58.65 7.43 5.99 H₂O 2.20

TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

Eluent: CH_2Cl_2 : MeOH: 25% NH_4OH (w/w) = 6:3:0.7 (v/v/v)

Detector: UV lamp (254 nm) or AcOH : H_2SO_4 conc. : p-

20 anisaldehyde $R_f = 0.24$

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

C) Title compound

According to the procedure described in EXAMPLE 1, 25 2 g of compound B) (2.17 mmol) in 45 ml of H₂O, were reacted with 0.80 g of GdCl₃ .6H₂O (2.17 mmol), maintaining at pH 6.5 by addition of 10.74 ml of 1N meglumine. 3.08 g of the desired product (2.1 mmol) were obtained.

30 Yield: 97%

45

Elemental analysis С Cl H Gđ N % calc.: 49.23 6.88 10.74 5.74 % found: 46.84 6.47

The IR and MS spectra are consistent with the indicated structure.

5.43 < 0.1

10.15

EXAMPLE 3

5

[[3,6,9-Tris(carboxymethyl)-10-(phenylmethoxy)methyl- $11-0x0-14-[[(3a,5\beta,7a,12a)-3,7,12-trihydroxy-24-oxo$ cholan-24-yl]amino]-3,6,9,12-tetraazatetradecanoate-

- 10 (4^{-})]gadolinate (1^{-})] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:1)
 - O-Phenylmethyl-N-[2-methoxy-2-oxoethyl]-N-[2-[[2- $\frac{1}{2}$ - $\frac{1}{$ [bis(2-methoxy-2-oxoethyl)amino]ethyl](2-methoxy-2-oxoethyl)amino]ethyl]-D,L-serine
- 15 A suspension of 40 g of 4-carboxy-5,8,11-tris-(carboxymethyl)-1-phenyl-2-oxa-5,8,11-triazatridecan-13-oic acid (prepared as described in EP-A-230893) (0.07789 mol) in 400 ml of anhydrous MeOH, kept at 0°C, was added with 150 ml of thionyl chloride, in 2 h. The 20 clear solution, heated to 25°C, was left under magnetic stirring for 30 h. The solution was evaporated to dryness and the resulting white solid, cooled in brine $(-15^{\circ}C)$, was added with 400 ml of Et₂O and, slowly and under stirring, with 500 ml of a $NaHCO_3$ saturated 25 solution (pH 10). After separation, the aqueous phase, kept at 0°C, was acidified to pH 6.5 with 6N HCl and then extracted with EtOAc. The organic phase evaporated to dryness. 23.4 g of the desired product (0.0411 mol) were obtained.
- 30 Yield: 53% HPLC titre: 98% (in % area)

46

Stationary phase: Column E. Merck Lichrosorb Select B, 5 µm; 250 x 4 mm;

Mobile phase: Gradient elution;

 $A = 0.01M \text{ KH}_2\text{PO}_4$ and 0.017M $H_3\text{PO}_4$ aqueous solution

5 $B = CH_3CN$

min	% A	% B
0	95	5
30	20	80
45	20	80

10 Flow rate: 1 ml min⁻¹;

Temperature: 45 °C;

Detector (UV): 210 nm, 254 nm and 280 nm.

TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

Eluent: CH_2Cl_2 : MeOH = 8:2 (v:v)

- Detector: 0.5% $\rm KMnO_4$ in 0.1N NaOH $\rm R_f$ = 0.5 The $^1\rm H-NMR$, $^{13}\rm C-NMR$, IR and MS spectra are consistent with the indicated structure.
- B) Methyl 3,6,9-tris(2-methoxy-2-oxoethyl)-10-(phe-nylmethoxy)methyl-11-oxo-14-[[(3α,5β,7α,12α)-3,7,-20 12-trihydroxy-24-oxocholan-24-yl]amino]-3,6,9,12-tetraazatetradecanoate

A solution of 39.2 g of compound A) (0.0688 mol), 33.1 g of (3α,5β,7α,12α)-N-(2-aminoethyl)-3,7,12-trihydroxycholan-24-amide (prepared according to the procedure described by Hilton, M.L.; Jones, A.S.; Westwood, J.R.B. J. Chem. Soc., 3449-3453, 1955) (0.0734 mol) and 13.33 g of diethoxyphosphoryl cyanide (0.076 mol) in 400 ml of DMF was added drop by drop, at 0°C and in 10 minutes, with 7.69 g of triethylamine (0.076 mol). After 4 h at 0°C and 16 h at room temperature, the reaction mixture was concentrated and

47

poured into a $NaHCO_3$ saturated solution and extracted with AcOEt. The organic phases were combined, washed with a NaCl saturated solution, with H_2O , dried over Na_2SO_4 and evaporated under reduced pressure. The solid residue was purified by flash chromatography to obtain 25.2 g of the desired product (0.0251 mol).

Yield: 36%

5

HPLC titre: 91% (in % area)

Stationary phase: Column E. Merck Lichrosorb Select

10 B, 5 μ m; 250 x 4 mm

Mobile phase: Gradient elution

 $A = 0.01M \text{ KH}_2\text{PO}_4$ and 0.017M H_3PO_4 aqueous solution

 $B = CH_3CN$

	min	% A	% B
15	o o	95	5
	. 30	20	80
	45	20	80

Flow rate:

 1 ml min^{-1}

Temperature:

30 °C

20 Detector (UV):

210 nm

TLC: Carrier: silica gel plates 60 F_{254} Merck

Eluent: CH_2Cl_2 : MeOH : 25% NH_4OH (w/w) = 9:1:0.1

(v/v/v)

Detector: AcCH : Conc. H_2SO_4 : p-anisaldehyde = 100:2:1

25 (v/v/v) R_f= 0.34

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

C) Title compound

11.5 g of compound B) (11.4 mmol) were dissolved in 1:1 MeOH/ $\mathrm{H}_2\mathrm{O}$ (300 ml) and 2N NaOH (5 ml) was added until pH 12 was reached. The reaction mixture was

stirred for 48 h at room temperature maintaining at pH 12 by addition of 1N NaOH (35 ml) through a pH-stat apparatus. The reaction was monitored by HPLC. resulting solution was adjusted to pH 7 with 2N HCl and evaporated. The residue was dissolved with 3:7 $MeOH/H_2O$ (500 ml), acidified with 6N HCl (15 ml) and the solution was loaded onto an Amberlite^R XAD-16 resin column and eluted with a $MeOH/H_2O$ gradient. Removal of the solvent from the fractions containing the product gave a solid that was further purified by reverse-phase 10 3,6,9-trispreparative HPLC to give 2.13 g of (carboxymethyl)-10-(phenylmethoxy)methyl-11-oxo-14- $[[(3a,5\beta,7a,12a)-3,7,12-trihydroxy-24-oxocholan-24-y1]$ amino]-3,6,9,12-tetraazatetradecanoic acid (2.25 mmol) as a white solid. The acid was suspended in H_2O (100 15 ml) and MeOH (20 ml) and 1N meglumine (5.6 ml; 5.6 mmol) was added until a complete dissolution (pH 6.8). A solution of $GdCl_3$ $6H_2O$ (0.83 g; 2.23 mmol) in H_2O (20 mL) was added drop by drop to the reaction mixture, maintained at pH 6.8 by addition of 1N meglumine (8.4 20 ml; 8.4 mmol). After 16 h the cloudy solution was filtered, loaded onto an Amberlite^R XAD-16 resin column and eluted with a $MeOH/H_2O$ gradient. The fractions containing the chelate were concentrated to dryness 25 under reduced pressure to give the desired product (2.0 g; 1.5 mmol).

Yield 13% m.p. = >300

K.F. titre: 4.56% (w/w)

HPLC titre: 98% (in % area)

30 Stationary phase: Column E. Merck Superspher RP-18; 250 x 4 mm

49

Mobile phase: Gradient elution

A = aq. 0.05 M KH₂PO₄

 $B = CH_3CN$

	min	% A	% B
5	0	70	30
	30	50	50

Flow rate:

 1 ml min^{-1}

Temperature: 40°C

Detector (UV): 210 nm

10 Elemental analysis

	С	H	N	Gđ		
% calc.:	50.99	6.92	6.48	12.14		
% found:	49.26	7.26	6.20	11.57	H ₂ O <	0.1
The IR	and MS	spectra	are co	nsistent	with	the
structure						

EXAMPLE 4

15

20

25

30

[[10-[2-0xo-2-[[3-[[2-[[(3α,5β,7α,12α)-3,7,12-trihydro-xy-24-oxocholan-24-yl]amino]ethyl]amino]propyl]amino]-ethyl]-1,4,7,10-tetraazacyclododecan-1,4,7-triacetoate-(3-)]gadolinate(0)] hydrogen compound with HCl (1:1)

A) 2-(2-Aminoethyl)-1,3-dioxolane

A suspension of 50 g of 2-(2-bromoethyl)-1,3-dioxolane (product known in literature, CAS RN = 5754-35-8) (0.27 mL, 32.5 mol), 62.5 g of potassium phthalimide (0.34 mol), 9.16 g of $\mathrm{Bu_4N^+HSO_4^-}$ (0.027 mol) in 150 ml of toluene was heated to 100°C and under $\mathrm{N_2}$ stream for 3 h. After cooling to room temperature, the mixture was filtered and evaporated to dryness. By crystallization of the residue from abs. EtOH the phthalimido derivative was obtained. A solution of 58.5 g of $\mathrm{NH_2NH_2.H_2O}$ (1.17 ml; 56.8 mol), 64.36 g of

PCT/EP95/01958 WO 95/32741

50

phthalimido derivative (0.26 mol) in 2 l of abs. EtOH was heated to reflux under N_2 stream for 2.5 h. After cooling to 0°C, the precipitated phthalhydrazide was filtered through a sintered funnel. By evaporation of the filtrate to dryness, 23.26 g of the desired product (0.198 mol) were obtained.

Yield: 73%

5

Elemental analysis C H N % calc.: 51.25 9.48 % found: 49.27 9.77 10 10.53 The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

- 2-[2-[(2-chloro-1-oxoethyl)amino]ethyl]-1,3-dioxo-
- A solution of 23.0 g of compound A) (0.196 mol) 15 and 19.8 g of Et_3N (0.196 mol; 27.16 ml) in 90 ml of CHCl3 under N2 stream was added with a solution of 22.17 g of chloroacetyl chloride (0.196 mol; 15.6 ml) in 60 ml of CHCl₃ keeping the temperature at 0-10°C.
- 20 When the reaction was completed, the reaction mixture was washed with H2O and the aqueous phase was extracted with CHCl3. The combined organic phases were dried and evaporated to dryness. By crystallization of the residue from Et₂O, 30.6 g of the desired product (0.158 25 mol) were obtained.

m.p.: 62-63°C (dec.) Yield: 81%

Elemental analysis С Cl H N 43.40 6.25 18.30 % calc.: 7.23 % found: 43.13 6.22 18.28

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent 30 with the indicated structure.

51

C) N-[2-(1,3-dioxolan-2-yl)ethyl]-1,4,7,10-tetraazacyclododecane-1-acetamide

solution of 203.3 g of 1,4,7,10tetraazacyclododecane (marketed product) (1.18 mol) in 2 l CH_3CN was added at 80°C and under N_2 stream with a solution of 23 g of compound B) (0.118 mol) in 500 ml of CH_3CN in 2 h. When the reaction was over, the reaction mixture was concentrated and the precipitate (1,4,7,10-tetraazacyclododecane excess) was off. The residue was evaporated to dryness and purified by column chromatography to obtain 36 g of the desired product (0.109 mol).

Yield: 93%

5

10

15

Elemental analysis C H N % calc.: 54.67 9.50 21.26 % found:

54.18

TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

 $CHCl_3 : MeOH : NH_4OH = 4:4:2$ Eluent:

Detector: UV lamp (254 nm) or ${\rm KMnO_4}$ in NaOH

20 $R_f = 0.3$

> The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

9.49

20.91

- 10-[2-oxo-2-[(3-oxopropyl)amino]ethyl]-1,4,7,10tetraazacyclododecane-1,4,7-triacetic acid
- 25 48.6 g of bromoacetic acid (0.35 mol) were dissolved in 40 ml of ${\rm H}_2{\rm O}$ and, keeping the temperature < 10°C, the pH of the solution was adjusted to 5 by addition of 175 ml of 2N NaOH. The solution, after addition of 35 g of compound C) (0.106 mol), was heated 30 to 50°C for 5 h, maintaining at pH 10 by addition of 160 ml of 2N NaOH (0.32 mol). The reaction mixture was

10

15

20

25

30

added with 30 ml of 37% HCl to pH 2 and the solution was heated for 2 h at 50°C. The reaction mixture was salted off by electrodialysis and after evaporating the aqueous solution and drying the residue, 40 g of the desired product (0.087 mol) were obtained.

Yield: 82% m.p.: 115-120°C

K.F. titre: 8.81% (w/w)

Elemental analysis C H N % calc.: 49.66 7.25 15.24 % found: 45.30 8.09 13.38

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

E) 10-[2-0x0-2-[[3-[[2-[[(3α,5β,7α,12α)-3,7,12-trihy-droxy-24-oxocholan-24-y1]amino]ethyl]amino]propyl]amino]ethyl]-1,4,7,10-tetraazacyclododecan-

1,4,7-triacetic acid

A solution of 80 g of N-(2-aminoethyl)-(3α,5β,-7α,12α)-3,7,12-trihydroxycholan-24-amide (prepared according to the procedure described by Hilton, M.L.; Jones, A.S.; Westwood, J.R.B., J. Chem. Soc. 1955, 3449-3453) (178 mmol) in 800 ml of anhydrous MeOH was added with 16.31 g of compound (D) (36 mmol), 35 ml of 1N HCl (35 mmol) and 1.49 g of NaBH₃CN (24 mmol). The solution was kept under nitrogen and magnetic stirring and in the presence of molecular sieves (0.4 nm). After 50 h the solvent was evaporated off under reduced pressure to obtain a crude that was purified by flash chromatography. The product was dissolved in 1N HCl and eluted through an Amberlite^R XAD-16 polystyrene resin, to obtain 8.79 g of the desired product (9.8 mmol).

Yield: 28% m.p.: 154°C

53

K.F. titre: 10.64% (w/w)

HPLC titre: 98.9% (in % area)

Stationary phase: Column E. Merck Lichrosorb Select

B; 5 μ m; 250 x 4 mm

5 Mobile phase: Gradient elution

A = 0.01M $\mathrm{KH_2PO_4}$ and 0.017M $\mathrm{H_3PO_4}$ aqueous solution

 $B = CH_3CN$

	min	% A	% B
	0	95	5
10	30	20	80
	45	20	80

Flow rate: 1 ml min^{-1}

Temperature: 45 °C

Detector (UV): 210 nm

15 Elemental analysis C H N % calc.: 60.44 8.91 10.97 % found: 53.72 9.49 9.51

TLC: Carrier: silica gel plates 60 F_{254} Merck

Eluent: CH_2Cl_2 : MeOH: 25% NH_4OH (w/w) = 7:3:1

 $20 \quad (v/v/v)$

Detector: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100:2:1 (v/v/v) $R_f = 0.26$

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

25 F) Title compound

A solution of 6.40 g of compound (E) (7.2 mmol) in 60 ml of $\rm H_2O$, kept at 50°C with stirring and under nitrogen atmosphere, was added with 1.18 g of $\rm GdO_3$ (3.3 mmol). pH before the addition of the oxide was 6.65.

30 After that, 1N HCl (7.2 ml) was added and pH was lowered to 2.75. The solution was filtered through a

Millipore filter (HAS 0.45 μm) and the solvent was evaporated off under reduced pressure, to obtain 6.90 g of the desired product (6.36 mmol).

Yield: 88.34% m.p.: 294°C (dec.)

- 5 Elemental analysis
 - C H N Gd Cl % calc.: 49.82 7.15 9.04 14.50 3.27 % found: 47.37 7.78 8.45 13.53 3.09

% found: 47.37 7.78 8.45 13.53 3.09 $\rm H_2O$ 6.44 TLC: Carrier: silica gel plates 60 $\rm F_{254}$ Merck

Eluent: $CHCl_3$: MeOH: H_2O : Et_3N = 8:2:0.1:0.1

Detector: AcOH: Conc. H_2SO_4 : p-anisaldehyde = 100:2:1 (v/v/v) R_f = 0.13

The IR and MS spectra are consistent with the indicated structure.

In the same way, the gadolinium complexes of the following ligands were prepared: $(3\beta, 5\beta, 7\alpha, 12\alpha) - 7, 12 - dihydroxy - 3 - [[[[3-[[[4,7,10-tris-$

(carboxymethyl)-1,4,7,10-tetraazacyclodec-1-yl]acetyl]-

amino]propyl]amino]acetyl]amino]-cholan-24-oic acid

- 20 (Compound 8);
 - (3β,5β,7α,12α)-7,12-dihydroxy-3-[[3-[[[4,7,10-tris(car-boxymethyl)-1,4,7,10-tetraazacyclododec-1-yl]acetyl]-amino]propyl]amino]-cholan-24-oic acid (Compound 21).

EXAMPLE 5

- [[(3β,5β,7α,12α)-3-[[13-carboxy-6,9,12-tris(carboxyme-thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-te-trazatridecyl]amino]-7,12-dihydroxy-cholan-24-oate-(5-)]gadolinate(2-)] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:2)
- 30 A) $(3\beta,5\beta,7\alpha,12\alpha)-3-azido-7,12-dihydroxy-cholan-24-$ oic acid methyl ester

A solution of 2.06 g of cholic acid methyl ester (marketed product) (4.87 mmol), 1.28 triphenylphosphine (4.88 mmol) and 1.70 of diethylazadicarboxylate (0.76 mL, 4.88 mmol) in 50 ml 5 of THF, at room temperature and under inert atmosphere, was added with а solution of g diphenylphosphorylazide (1.1 mL, 5.11 mmol) in 5 ml of during 15 minutes. After 24 hours at temperature, 1 equivalent of diethylazadicarboxylate 10 (0.76)mL, 4.88 mmol) and 1 equivalent triphenylphosphine (1.28 g, 4.88 mmol) were added. After a further 24 hours the solvent was evaporated off under reduced pressure and the resulting crude was purified by flash chromatography. 1.6 g of the desired 15 product (3.57 mmol) was obtained.

Yield: 73%

25

Elemental analysis C H N

% calc.: 67.08 9.23 9.38

% found: 66.86 9.30 9.15 H₂O 0.74

20 TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

Eluent: AcOEt : hexane = 1:1

Detector: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100:2:1

 $(v/v/v) R_f = 0.56$

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

B) $(3\beta,5\beta,7\alpha,12\alpha)-3$ -amino-7,12-dihydroxy-cholan-24-oic acid methyl ester

A solution of 28.28 g of compound A) ester methyl (0.063 mol) in 100 ml of THF was added with 5 ml of H_2O and 16.59 g of triphenylphosphine (0.063 mol). After 96 h at room temperature, the reaction mixture was

WO 95/32741

56

evaporated under reduced pressure and the residue was purified by flash chromatography, to obtain 23.21 g of the desired product (0.055 mol).

Yield 87%.

5 Elemental analysis C H N

> 71.29 % calc.: 10.39 3.32

% found: 70.06 10.57 3.41 H₂O 0.26

TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

MeOH : $Et_3N = 95:5$ Eluent:

Detector: AcOH : Conc. H₂SO₄ : p-anisaldehyde = 100:2:1 10 (v/v/v) $R_f = 0.36$

The $^{1}H-NMR$, $^{13}C-NMR$, IR and MS spectra are consistent with the indicated structure.

 $(3\beta,5\beta,7\alpha,12\alpha)-3-[[[(phenylmethoxy)carbonyl]ami-$ 15 no]acetyl]amino]-7,12-dihydroxy-cholan-24-oic acid methyl ester

A solution of 12.25 g of carbobenzyloxyglycine (marketed product) (58.5 mmol) and 6 g of N-methylmorpholine (59.3 mmol) in 400 ml of THF was added drop by drop, under nitrogen and at -4°C, with 8 g of 20 isobutyl chloroformate (58.4 mmol) and subsequently 21.8 g of compound B) (51.7 mmol) dissolved in 100 ml of THF. After 30 min at -4°C the reaction mixture was filtered and evaporated under reduced pressure. 25 residue was taken up into Et₂O and H₂O; the organic phase was separated, washed with H_2O , dried over Na_2SO_4 and evaporated under reduced pressure. The residue was purified by flash chromatography, to obtain 27.9 g of the desired product (45.5 mmol).

Yield: 88%. 30

57

Elemental analysis C H N

% calc.: 68.59 8.55 4.57

% found: 68.27 8.74 4.52 H₂O 0.25

TLC: Carrier: silica gel plates 60 F_{254} Merck

5 Eluent: MeOH: $Et_3N = 95:5$

Detector: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100:2:1

 $(v/v/v) R_f = 0.85$

15

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

D) (3β,5β,7α,12α)-3-(aminoacetyl)amino-7,12-dihydro-xy-cholan-24-oic acid methyl ester

A solution of compound C) methyl ester in MeOH was added with 10% Pd/C and the mixture was hydrogenated at room temperature and pressure, to obtain the desired product.

- E) (3β,5β,7α,12α)-3-[[13-carboxy-6,9,12-tris(carboxy-methyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,-9,12-tetraaazatridecyl]amino]-7,12-diidroxy-cholan-24-oic acid
- 20 According to the procedure described in EXAMPLE 3, O-phenylmethyl-N-[2-methoxy-2-oxoethyl]-N-[2-[[2-[bis-(2-methoxy-2-oxoethyl)amino]ethyl](2-methoxy-2-oxoethyl)amino]ethyl]-D,L-serine and compound D) were condensed, in DMF triethylamine, and 25 diethoxyphosphoryl cyanide. When the reaction was over, the reaction mixture was poured into a $NaHCO_3$ saturated solution and extracted with AcOEt. The organic phases were combined and evaporated under reduced pressure. The residue was dissolved in MeOH and hydrolysed with a 30 LiOH monohydrate aqueous solution. The reaction mixture was evaporated to dryness, the solid residue

58

dissolved in 1N HCl and eluted through an Amberlite $^{\rm R}$ XAD-16 polystyrene resin, to obtain the desired product.

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

F) Title compound

5

10

According to the procedure described in EXAMPLE 1, compound E) was reacted with $GdCl_3.6H_2O$ in H_2O , maintaining at pH 6.5 by addition of 1N meglumine. The desired product was obtained.

The IR and MS spectra are consistent with the indicated structure.

In the same way, the gadolinium complexes of the following ligands were prepared:

- 3,6,9-Tris(carboxymethyl)-14-[[(3β,5β,7α,12α)-7,12-di-hydroxy-24-oxo-24-[(2-sulfoethyl)amino]cholan-3-yl]ami-no]-11,14-dioxo-10-(phenylmethoxy)methyl-3,6,9,12-te-traazatetradecanoic acid (COMPOUND 15);
 - $N-[(3\beta,5\beta,7\alpha,12\alpha)-3-[[13-carboxy-6,9,12-tris(carboxyme-$
- 20 thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-tetraazatridecyl]amino]-7,12-dihydroxy-24-oxocholan-24yl]glycine (COMPOUND 16);
 - $(3\beta,5\beta,7\alpha)-3-[[13-carboxy-6,9,12-tris(carboxymethyl)-$
 - 1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-tetraaza-
- tridecyl]amino]-7-hydroxy-cholan-24-oic acid (COMPOUND
 17);
 - $(3\beta,5\beta,12\alpha)-3-[[13-carboxy-6,9,12-tris(carboxymethyl)-$
 - 1,4-dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-tetraaza-
 - tridecyl]amino]-12-hydroxy-cholan-24-oic acid (COMPOUND
- 30 18);
 - $(3\beta, 5\beta)-3-[[13-carboxy-6, 9, 12-tris(carboxymethyl)-1, 4-$

dioxo-5-[(phenylmethoxy)methyl]-3,6,9,12-tetraazatride-cyl]amino]-cholan-24-oic acid (COMPOUND 19); (3β,5β,7α,12α)-3-[[17-carboxy-10,13,16-tris(carboxyme-thyl)-1,8-dioxo-9-[(phenylmethoxy)methyl]-7,10,13,16-tetraazaheptadecyl]amino]-7,12-dihydroxy-cholan-24-oic acid (COMPOUND 20).

EXAMPLE 6

[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[17-carboxy-10,13,16-tris(carboxy-methyl)-8-oxo-9-[(phenylmethoxy)methyl]-3,7,10,13,16-

- pentaazaheptadecyl]oxy]-7,12-dihydroxy-cholan-24-oate-(5-)]gadolinate(2-)] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol(1:2)
 - A) 2-Chloro-N-[2-(1,3-dioxolan-2-yl)ethyl]-3-phenyl-methoxypropanamide
- 15 A solution of 69.63 g of 2-chloro-3-(phenylmethoxy)propanoyl chloride (prepared according to the procedure described in Inorg. Chem., 31. 2422, 1992) (0.299 mol) in 90 ml of CHCl₃ was added to a solution of 35.49 g of 2-(2-aminoethyl)-1,3-dioxolane (prepared 20 according to the procedure described in EXAMPLE 4) (0.303 mol) and to 60.3 g of triethylamine (83 ml; 0.596 mol) in 100 ml of CHCl_3 under inert atmosphere, keeping the temperature at 0-5°C. The reaction mixture was stirred for 5 h at 25°C, then was washed with ${\rm H}_2{\rm O}$. The organic phase was dried and evaporated to dryness, 25 the residue was purified by flash chromatography. 61.68 g of the desired product (0.197 mol) were obtained.

Yield: 66%

TLC: Carrier: silica gel plates 60 F_{254} Merck

30 Eluent: AcOEt : n-hexane = 1 : 1 (v/v)

Detector: 0.5% $KMnO_4$ in 0.1N NaOH $R_f = 0.34$

10

15

20

25

30

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

B) 5,8,11-tris[2-(1,1-dimethylethoxy)-2-oxoethyl]-1phenyl-4-[[2-(1,3-dioxolan-2-yl)ethyl]amino]carbonyl-2-oxa-5,8,11-triazatridecan-13-oic acid (1,1dimethylethyl) ester

30.97 g of diethylenetriamine (0.300 mol) were added to a solution of 20.94 g of compound A) (0.067 mol) in 100 ml of MeCN under inert atmosphere and the mixture was kept at 50°C for 72 h and at 80°C for 8 h. After cooling to 0°C, the precipitate (diethylenetriamine hydrochloride) was filtered and washed with 50 ml of MeCN. After evaporating the solvent under reduced pressure, the diethylenetriamine excess was distilled off under vacuum. The crude was taken up into 80 ml of AcOEt, filtered and evaporated to dryness to obtain a residue, that was purified by column chromatography [silica gel; eluent $CHCl_3$: MeOH: NH_3 25% (w/w) = 20:4:0.4 (v/v/v). 12.61 g of 2-[[2-[(2-aminoethyl)amino]ethyl]amino]-3-(phenylmethoxy)-N-[2-(1,3-dioxolan-2y1)ethy1]propanamide (0.03 mol) were obtained, that was used directly in the subsequent step (46% yield).

A solution of 7.50 g of 2-[[2-[(2-aminoethyl)amino]ethyl]amino]-3-(phenylmethoxy)-N-[2-(1,3-dioxolan-2-yl)ethyl]propanamide in 30 ml of 1,2-dichloroethane was added, under inert atmosphere, with 20.64 g of diisopropylethylamine (0.160 mol) and, keeping the temperature from 0 to 5°C, with 15.58 g of t-butyl bromoacetate (0.080 mol). The solution was kept at 15°C for 24 h, added with further t-butyl bromoacetate (4.25 g; 0.022 mol) and kept for 72 h at 15°C. The solution

was cooled to 0°C and filtered. The filtrate was concentrated, taken up into $\rm H_2O$ and extracted with AcOEt. The organic phase was washed with $\rm H_2O$, dried and evaporated to dryness to obtain a crude that was purified by column chromatography [silica gel 935 g; eluent: AcOEt: n-hexane 1:1 (v/v)]. The fractions of similar purity were combined and evaporated to dryness to obtain the desired product (5.18 g; 0.062 mmol). Yield: 34%.

10 Yield: 16% on two steps

15

TLC: Carrier: silica gel plates 60 F_{254} Merck

Eluent: AcOEt : n-hexane = 1 :1

- 5,8,11-tris(carboxymethyl)-1-phenyl-4-[(3-oxopro-pyl)amino]carbonyl-2-oxa-5,8,11-triazatridecan-13-oic acid
- 67 ml of 1N HCl (0.067 mol) were added to a 20 solution of 14 g of compound B) (0.017 mol) in 280 ml of dioxane. The solution was diluted with 215 ml of $\rm H_2O$, stirred at 35°C for 54 h, then at 4°C for 48 h. After evaporation of the dioxane, the aqueous solution was extracted with AcOEt. The organic phase was washed with $\mathrm{H}_2\mathrm{O}$, then dried and evaporated to dryness. The residue was taken up into $\mathrm{CH}_2\mathrm{Cl}_2$ and the solution was evaporated to dryness. The residue was taken up into $\mathrm{CH_{2}Cl_{2}}$ and the solution was added, in about 1 h, with 82 g of trifluoroacetic acid (55.7 ml; 0.719 mol). The 30 solution was kept at 5°C for 24 h under atmosphere, then was evaporated to dryness. The residue

was taken up into $\mathrm{CH_2Cl_2}$ and evaporated to dryness, repeating the procedure several times. The crude was taken up into $\mathrm{CH_2Cl_2}$ and extracted with $\mathrm{H_2O}$. The aqueous phase was separated, evaporated to small volume and chromatographed by HPLC. 1.5 g of the desired product (2.64 mmol) were obtained.

Yield: 16% m.p.: 100-102°C (dec.)

K.F. titre: 2.27% (w/w)

HPLC titre: 97% (in % area)

10 Stationary phase: Column E. Merck Lichrosorb RP-Select B 5 µm; 250 x 4 mm;

Mobile phase: Gradient elution;

 $A = 0.01M \text{ KH}_2PO_4$ and $0.017M \text{ H}_3PO_4$ aqueous solution

 $B = A / CH_3CN = 3:7$

15	min	% A	% B
	0	90	10
	30	10	90
	40	10	90

Flow rate: 1.5 ml min^{-1} ;

20 Temperature: 35 °C;

Detector (UV): 210 nm.

Elemental analysis

C H N Na Cl H₂O % calc.: 52.81 6.38 9.85 % found: 51.82 6.34 9.62 < 0.10 < 0.10 2.27

25 % found: 51.82 6.34 9.62 < 0.10 < 0.10 2.27 The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

- D) (3β,5β,7α,12α)-3-[[17-carboxy-10,13,16-tris(carbo-xymethyl)-8-oxo-9-[(phenylmethoxy)methyl]-3,7,10,-
- 30 13,16-pentaazaheptadecyl]oxy]-7,12-dihydroxycho-lan-24-oic acid

63

According to the procedure described in EXAMPLE 4, compound C) and $(3\beta,5\beta,7\alpha,12\alpha)-3-[2-(amino)ethoxy]-7,12-dihydroxycholan-24-oic acid (prepared according to the procedure described in EP-A-417725), were reacted, in anhydrous MeOH and HCl, with NaBH₃CN, under inert atmosphere. The desired product was obtained. The <math>^{1}$ H-NMR, 13 C-NMR, IR and MS spectra are consistent with the indicated structure.

E) Title compound

10 According to the procedure described in EXAMPLE 1, compound D) was reacted with $GdCl_3.6H_2O$ in H_2O , maintaining at pH 6.5 by addition of 1N meglumine. The desired product was obtained.

The IR and MS spectra are consistent with the indicated structure.

EXAMPLE 7

5

15

20

25

30

[[(3\beta,5\beta,7\alpha,12\alpha)-7,12-dihydroxy-3-[2-[[[[4-[4,12-bis-(carboxy)-5,8,11-tris(carboxymethyl)-2-oxa-5,8,11-tria-zadodecyl]phenyl]amino]thioxomethyl]amino]ethoxy]-cho-lan-24-oate(6-)]gadolinate(3-)] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:3)

A) (3β,5β,7α,12α)-7,12-dihydroxy-3-[2-[[[[4-[4,12-bis(carboxy)-5,8,11-tris(carboxymethyl)-2-oxa-5,8,11-triaazadodecyl]phenyl]amino]thioxomethyl]-amino]ethoxy]-cholan-24-oic acid

A solution of 4-[(1,1-dimethylethoxy)carbonyl]-5,8,11-tris[2-(1,1-dimethylethoxy)-2-oxoethyl]-1-(4-aminophenyl)-2-oxa-5,8,11-triazatridecan-13-oic acid 1,1-dimethylethyl ester (prepared according to the procedure described in EXAMPLE 1) in CHCl₃ was added with 1,1'-thiocarbonyl diimidazole (marketed product)

15

and subsequently with $(3\beta,5\beta,7\alpha,12\alpha)-3-[2-(amino))$ etho-xy]-7,12-dihydroxy-cholan-24-oic acid (prepared according to the procedure described in EP-A-417725). The reaction mixture was then evaporated and the residue dissolved in CH₂Cl₂ and hydrolysed with CF₃COOH to give the desired product.

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

- B) Title compound
- 10 According to the procedure described in EXAMPLE 1, compound A) was reacted with $GdCl_3.6H_2O$ in H_2O , maintaining at pH 6.5 by addition of 1N meglumine. The desired product was obtained.

The IR and MS spectra are consistent with the indicated structure.

EXAMPLE 8

[[3,6,9-Tris(carboxymethyl)-10-[(phenylmethoxy)methyl]- $11-0x0-17-[[(3\alpha,5\beta,7\alpha,12\alpha)-3,7,12-trihydroxy-24-oxocho-lan-24-yl]amino]-3,6,9,12-tetraazaoctadecanedioate-$

- 20 (5⁻)]gadolinate(2⁻)] dihydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:2)
 - A) N^6 -(phenylmethoxy)carbonyl- N^2 -[(3a,5 β ,7a,12a)-3,-7,12-trihydroxy-24-oxocholan-24-yl]-L-lysine methyl ester
- 25 A suspension of cholic acid (16.3 g; 40 mmol) and triethylamine (4.86 g; 48 mmol) in THF (350 ml), kept at 0°C under nitrogen atmosphere, was added drop by drop with isobutyl chloroformate (6.56 g; 48 mmol) in 10 min. After 30 Nemin a solution of 30 (phenylmethoxy)carbonyl-L-lysine (marketed product) (11.2 g; 40 mmol) in 0.67 N NaOH (60 ml) was dropped

10

therein during 20 min. The reaction mixture was kept at 0°C for one more hour and then at room temperature for 5 h. A 2N HCl aqueous solution was added to the mixture until acid pH, then the organic solvent was evaporated off under reduced pressure. The residual aqueous suspension was diluted with a NaCl saturated solution and extracted with AcOEt. The organic phases were combined, dried and evaporated to dryness, recovering a solid that was powdered and dried over P_2O_5 under reduced pressure. The resulting crude product was subjected to the subsequent reaction, without further purification procedures.

A solution of N^6 -(phenylmethoxy)carbonyl- N^2 -[(3a,5 β ,7a,12a)-3,7,12-trihydroxy-24-oxocholan-24-yl]-

L-lysine (27.5 g) in MeOH (600 ml), kept at room temperature and under nitrogen atmosphere, was added with p-toluenesulfonic acid monohydrate (1.56 g; 8.2 mmol). After 20 h the reaction mixture was added with Et₃N (0.832 g; 8.2 mmol). The mixture was evaporated under reduced pressure and the resulting crude was purified by flash chromatography to obtain the desired

Yield: 88% m.p.: 80-83°C

K.F. titre: 1.37% (w/w)

product (24.1 g; 35.2 mmol).

25 HPLC titre: 99.5% (in % area)

Stationary phase: Column E. Merck Lichrosorb Select

B; 5 μ m; 250 \times 4 mm

Mobile phase: Gradient elution

 $A = 0.01M \text{ KH}_2\text{PO}_4$ and 0.017M $H_3\text{PO}_4$ aqueous solution

 $B = CH_3CN$

PCT/EP95/01958 WO 95/32741

66

min	% A	% B
0	95	5
30	20	80
45	20	80

Flow rate: 1 ml min $^{-1}$ 5

Temperature: 45 °C

Elemental analysis

Detector (UV): 210 nm $[a]^{20}_{D}$: + 16.2° (c 2.1; MeOH)

10 % calc.: 68.39 8.83 4.09 % found: 66.82 9.01

TLC: Carrier: silica gel plates 60 F₂₅₄ Merck

Eluent: AcOEt : i-PrOH = 9:1 (v/v)

Detector: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100:2:1

H

N

15 (v/v/v) $R_{f} = 0.22$

> The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

- B) $N^2-[(3\alpha,5\beta,7\alpha,12\alpha)-3,7,12-\text{trihydroxy}-24-\text{oxocholan}-$ 24-yl]-L-lysine methyl ester monohydrochloride
- 20 A solution of 15.0 g of compound A) (21.9 mmol) in MeOH (150 ml) was added with Pd/C (1.5 g). The mixture was hydrogenated at room temperature and pressure. The transformation was monitored by TLC and HPLC. After 1.5 h the reaction mixture was filtered through paper 25 filter and the filtrate was cooled to $0^{\circ}C$ in a $H_{2}O/ice$ bath and added with a HC solution in MeOH (20.5 ml; 23.2 mmol). The solution was evaporated under reduced pressure and the residue was powdered and dried under reduced pressure to obtain the desired product (12.4 g;

21.1 mmol).

30

Yield: 96% m.p.: 108-110°C WO 95/32741

67

K.F. titre: 2.20% (w/w)

HPLC titre: 92.4% (in % area)

Stationary phase: Column E. Merck Lichrosorb Select

B; 5 μ m; 250 \times 4 mm

5 Mobile phase: Gradient elution

A = 0.01M $\mathrm{KH_2PO_4}$ and 0.017M $\mathrm{H_3PO_4}$ aqueous solution

 $B = CH_3CN$

	min	% A	% B
	0	95	5
10	30	20	80
	45	20	80

Flow rate: 1 ml min^{-1}

Temperature: 45 °C

Detector (UV): 210 nm

15 $[\alpha]^{D}_{20} = +14.4^{\circ} (c 2.16, MeOH)$

Elemental analysis C H N Cl % calc.: 63.40 9.44 4.77 6.04

% found: 62.01 9.89 4.59 5.93

TLC: Carrier: silica gel plates 60 F_{254} Merck

20 Eluent: MeOH : $Et_3N = 95:5 (v/v)$

Detector: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100:2:1

 $(v/v/v) R_f = 0.33$

The $^{1}\text{H-NMR}$, $_{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

- 25 C) 3,6,9-Tris(carboxymethyl)-10-[(phenylmethoxy)me-thyl]-11-oxo-17-[[(3α,5β,7α,12α)-3,7,12-trihydro-xy-24-oxocholan-24-yl]amino]-3,6,9,12-tetraaza-octadecanedioic acid
- 9.35 g of compound B) (14.3 mmol), O-phenylmethylN-(2-methoxy-2-oxoethyl)-N-[2-[[2-[bis(2-methoxy-2-oxo-ethyl)amino]ethyl](2-methoxy-2-oxoethyl)amino]ethyl]-

D,L-serine (prepared as described in Example 3) (9.28 g; 14.3 mmol) and BOP-reagent (marketed product) (6.32 g; 14.3 mmol) were dissolved in DMF (140 ml) at room temperature. N, N-Diisopropylethylamine (8.51 ml; 50.1 mmol) was added to this stirred solution in 15 min. The reaction was monitored by HPLC. After 6 h the reaction mixture was evaporated under reduced pressure and the residue was dissolved in EtOAc. The solution was successively washed with saturated aqueous NH₄Cl, H₂O up to neutral pH, dried and evaporated under reduced 10 pressure. The solid residue was purified by flash chromatography to obtain a yellow-brown solid that was dissolved in 2:1 MeOH/ H_2O and 1N NaOH (1.5 ml) was added until pH 12 was reached. The reaction mixture was stirred for 21 h at room temperature maintaining at pH 15 12 by addition of 1N NaOH (35.5 ml) through a pH-stat apparatus. The reaction was monitored by HPLC. The resulting solution was adjusted to pH 6.5 with 1N HCl and evaporated under reduced pressure. The residue was dissolved in 7:3 1N HCl/MeOH and the solution was 20 loaded onto an Amberlite^R XAD-16.00 resin column and eluted with a MeOH/H2O gradient. The fractions containing the ligand were concentrated to dryness under reduced pressure giving the desired product (4.51 25 g; 4.37 mmol).

Yield: 31% m.p.: 158-160°C

K.F.: 4.13% (w/w)

HPLC: 97% (area %)

Stationary phase: Column E. Merck Lichrosorb Select

30 B; 5 μm; 250 x 4 mm

Mobile phase: Gradient elution

69

A = aqueous solution 0.01 M in KH_2PO_4 and 0.017 M in H_3PO_4

 $B = CH_3CN$

	min	% A	% B
5	0	95	5
	30	20	80
	45	20	80

Flow rate: 1 ml min^{-1}

Temperature: 45 °C

10 Detector (UV): 210 nm

Elemental analysis C H N Cl % calc.: 60.50 7.91 6.79 0.00 % found: 58.13 8.22 6.33 <0.1

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719)

Eluent: $80:30:5:5 = CHCl_3 : MeOH : H_2O : Et_3N$ Detection: AcOH : Conc. $H_2SO_4 : p$ -anisaldehyde = $100 : 2 : 1 (v/v/v) R_f = 0.25$ The 1H -NMR, ^{13}C -NMR, IR and MS spectra are consistent with the structure.

20 D) Title compound

25

30

3.52 g of compound C) (3.20 mmol) were suspended in 9: 1 $\rm H_2O/MeOH$ (70 ml) at 50°C and under nitrogen. Meglumine (1.241 g; 6.357 mmol) was added obtaining complete dissolution. $\rm Gd2O_3$ (0.581 g; 1.60 mmol) was added to the reaction mixture and the resulting suspension was stirred for 21 h at 50°C. The almost clear solution was filtered through Millipore^R apparatus (HAS 0.45 μ m filter) and the filtrate was adjusted to pH 7 with 1% meglumine solution (0.90 ml; 9.0 mg; 4.6 μ mol). The solution was evaporated to dryness under reduced pressure to obtain a solid that

70

was pulverized and dried under reduced pressure giving the desired product (4.90 g; 3.11 mmol).

Yield 94% m.p. 170 - 175°C (160°C, sint.)

K.F. titre: 2.15% (w/w)

5 HPLC titre: 97% (area %)

Stationary phase: Column E. Merck Superspher RP-18;

250 x 4 mm

Mobile phase: Isocratic elution: 74 : 26 A/B

A = 0.05 M aqueous solution in KH_2PO_4

 $B = CH_3CN$

Flow rate: 1 ml min^{-1}

Temperature: 45°C

Detector (UV): 210 nm

Elemental analysis C H N Gd

15 % calc.: 50.27 7.16 6.22 9.97

% found: 49.77 7.49 6.07 9.68

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719)

Eluent: 80 : 30 : 5 : 5 $CHCl_3$: MeOH : H_2O : Et_3N

Detection: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100 :

20 2:1 (v/v/v) R_f= 0.33

The IR and MS spectra are consistent with the structure.

EXAMPLE 9

 $[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[13-carboxy-6,9,12-tris(carboxyme-$

- 25 thyl)-1,4-dioxo-3,6,9,12-tetraaazatridecyl]amino]-7,12dihydroxy-cholan-24-oate(5-)]gadolinate(1-)] hydrogen
 compound with 1-deoxy-1-(methylamino)-D-glucitol (1:2)
 - A) (3β,5β,7α,12α)-3-[[13-Carboxy-6,9,12-tris(carboxy-methyl)-1,4-dioxo-3,6,9,12-tetraaazatridecyl]ami-
- no]-7,12-dihydroxy-cholan-24-oic acid

A suspension of diethyleneetriaminepentaacetic

dianhydride (marketed product) (0.142 mol) in DMF (850 ml) at 80°C, was added drop by drop with a solution of H₂O (0.211 mol) and DMF (50 ml). After 1.5 h. a solution of (3β,5β,7α,12α)-3-(aminoacetyl)amino-7,12-dihydroxy-cholan-24-oic acid methyl ester (prepared according to the procedure of EXAMPLE 5) (0.0356 mol) in DMF (100 ml) was dropped therein. When the addition was over, the mixture was cooled to 20°C and 2N NaOH (360 ml) was dropped therein. After 24 h the mixture was adjusted to pH 7 with 37% HCl and the solution was evaporated under vacuum. The residue was dissolved with MeOH/H₂O=3/7 (500 ml) and with 37% HCl (7 ml). The resulting solution was loaded on an Amberlite^R XAD-16 resin and eluted with a MeOH/H₂O gradient to obtain the desired product.

The $_1^{\rm H-NMR}$, $^{13}{\rm C-NMR}$, IR and MS spectra are consistent with the indicated structure.

B) Title compound

5

10

15

According to the procedure described in EXAMPLE 1, compound A) was reacted with GdCl₃.6H₂O in H₂O, maintaining at pH 6.5 by addition of 1N meglumine. The desired product was obtained.

The IR and MS spectra are consistent with the indicated structure.

In the same way, the gadolinium complexes of the following ligands were prepared:

3,6,9-Tris(carboxymethyl)-14-[[(3 β ,5 β ,7 α ,12 α)-7,12-dihydroxy-24-oxo-24-[(2-sulfoethyl)amino]-cholan-3-yl]amino]-11,14-dioxo-3,6,9,12-tetraazatetradecanoic

30 acid (COMPOUND 22); [(3β,5β,7α,12α)-3-[[17-carboxy-10,13,16-tris(carboxyme-

10

15

20

25

thyl)-1,8-dioxo-7,10,13,16-tetraazaheptadecyl]amino]-7,12-dihydroxy-cholan-24-oic acid (COMPOUND 23); (17S)-3,6,9-Tris(carboxymethyl)-11-oxo-17-[[(3β,5β,7α,-12α)-3,7,12-trihydroxy-24-oxocholan-24-yl]amino]-3,6,-9,12-tetraazaoctadecanedioic acid (COMPOUND 24).

EXAMPLE 10

[$(3\beta,5\beta,7\alpha,12\alpha)$ - $(3'\beta,5'\beta,7'\alpha,12'\alpha)$ -3,3'-[[6,9,12-tris-(carboxymethyl)-1,4,14,17-tetraoxo-3,6,9,12,15-penta-azaheptadecan-1,17-diyl]bisimino]bis[7,12-dihydroxycho-lan-24-oate(5-)]gadolinate(2-)] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:2)

A) (3β,5β,7α,12α)-(3'β,5'β,7'α,12'α)-3,3'-[[6,9,12-tris(carboxymethyl)-1,4,14,17-tetraoxo-3,6,9,12,-15-pentaazaheptadecan-1,17-diyl]bisimino]bis[7,12-dihydroxy-cholan-24-oic] acid

Diethylenetriamino pentaacetic acid dianhydride (marketed product) was reacted in DMF with t.wo equivalents of $(3\beta, 5\beta, 7\alpha, 12\alpha)-3-(aminoacetyl)amino-$ 7,12-dihydroxycholan-24-oic acid methyl ester (prepared according to the procedure described in Example 5). The reaction mixture was subsequently treated with a LiOH monohydrate aqueous solution, evaporated and residue was dissolved in 1N HCl and eluted through an Amberlite^R XAD-16 polystyrene resin, to obtain the desired product.

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

B) Title compound

According to the procedure described in EXAMPLE 1, compound A) was reacted with $GdCl_3.6H_2O$ in H_2O , maintaining at pH 6.5 by addition of 1N meglumine. The

73

desired product was obtained.

The IR and MS spectra are consistent with the indicated structure.

EXAMPLE 11

25

30

- [[[3β(S),5β,7α,12α]-7,12-dihydroxy-3-[[4-[[[[5-[bis[2-[bis(carboxymethyl)amino]ethyl]amino]-5-carboxypentyl]amino]thioxomethyl]amino]benzoyl]amino]-cholan-24-oate(6-)]gadolinate(3-)] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:3)
- A) N²-Bis[2-[bis(carboxymethyl)amino]ethyl]-L-lysine
 This product was synthesized starting from N⁶(phenylmethoxy)carbonyl-L-lysine (marketed product)
 analogously to what described by M. A. Williams and H.
 Rapoport, J. Org. Chem. 1993, 58, 1151-1158 for the 4nitro-L-phenylalanine.

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

B) [3β(S),5β,7α,12α]-7,12-dihydroxy-3-[[4-[[[[5-[bis-[2-[bis(carboxymethyl)amino]ethyl]amino]-5-carbo-xypentyl]amino]thioxomethyl]amino]benzoyl]amino]-cholan-24-oic acid

solution of $(3\beta, 5\beta, 7\alpha, 12\alpha) - 3 - amino - 7, 12$ dihydroxycholan-24-oic acid (prepared according to the procedure described in EP-A-417725) in DMF and triethylamine was added with an equimolecular amount of 4-isothiocyanatobenzoyl chloride (prepared according to the procedure described by N. Viswanathan and R. C. Desai, Indian J. Chem., 1981. 20B, 308-310). After $(3\beta,5\beta,7\alpha,12\alpha)-3$ -amino-7,12-dihydroxycholan-24-oic acid had been completely converted, compound A) was added to the reaction mixture. When the reaction was over, the

25

30

solvent was evaporated off and the residue was dissolved in 1N HCl and eluted through an $Amberlite^R$ XAD-16 polystyrene resin, to obtain the desired product.

- 5 The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.
 - C) Title compound

According to the procedure described in EXAMPLE 1, compound B) was reacted with $GdCl_3.6H_2O$ in H_2O , maintaining at pH 6.5 by addition of 1N meglumine. The desired product was obtained.

The IR and MS spectra are consistent with the indicated structure.

EXAMPLE 12

- [[[3\beta(S),5\beta,7\alpha,12\alpha]-7,12-dihydroxy-3-[[4-[[5-[bis[2-[bis(carboxymethyl)amino]ethyl]amino]-5-carboxypentyl]-amino]-1,4-dioxobutyl]amino]-cholan-24-oate(6-)]gadolinate(3-)] hydrogen compound with 1-deoxy-1-(methyl-amino)-D-glucitol (1:3)
- 20 A) (3β,5β,7α,12α)-3-[(3-carboxy-1-oxopropyl)amino]7,12-dihydroxycholan-24-oic acid methyl ester

A solution of 6.15 g of $(3\beta,5\beta,7\alpha,12\alpha)-3$ -amino-7,12-dihydroxycholan-24-oic acid methyl ester (prepared according to the procedure described in EXAMPLE 5) (15 mmol) in 85 ml of THF and 17 ml of triethylamine was added with 1.5 g of succinic anhydride (15 mmol). After 4 h at room temperature the reaction mixture was poured into 200 ml of 1N HCl and extracted with AcOEt. The organic phase was washed with H_2O , dried and evaporated under reduced pressure. The residue was purified by flash chromatography, to obtain 4.5 g of the desired

75

product (8.6 mmol).

Yield: 57% m.p.: 92-94°C

Elemental analysis C H N

% calc.: 66.76 9.08 2.68

5 % found: 65.45 9.40 2.50 0.56 H₂O

TLC: Carrier: silica gel plates 60 F_{254} Merck

Eluent: AcOEt : AcOH = 4:1

Detector: AcOH : Conc. H_2SO_4 : p-anisaldehyde

 $R_{f} = 0.47$

30

- 10 The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.
 - B) (3β,5β,7α,12α)-3-[[4-[(2,5-dioxo-1-pyrrolidinyl)oxy]-1,4-dioxobutyl]amino]-7,12-dihydroxy-cholan-24-oic acid methyl ester
- A solution of compound A) in anhydrous THF and anhydrous acetonitrile was added with N-hydroxysuccinimide and subsequently with dicyclohexyllcarbodiimide: dicyclohexylurea precipitated and was filtered off. The solution was evaporated under reduced pressure to obtain the desired product.

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

- C) [3β(S),5β,7α,12α]-7,12-dihydroxy-3-[[4-[[5-[bis[2-[bis(carboxymethyl)amino]ethyl]amino]-5-carboxy-
- pentyl]amino]-1,4-dioxobutyl]amino]-cholan-24-oic
 acid

A solution of compound B) in DMF was added with a solution of N^2 -bis[2-[bis(carboxymethyl)amino]ethyl]-L-lysine (prepared according to the procedure described in Example 11) in DMF and triethylamine. After 24 h, the reaction mixture was added with a LiOH monohydrate

PCT/EP95/01958

WO 95/32741

76

aqueous solution, then the solvent was evaporated and the residue was dissolved in 1N HCl and eluted through an Amberlite R XAD-16 polystyrene resin, to obtain the desired product.

- The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent 5 with the indicated structure.
 - D) Title compound

According to the procedure described in EXAMPLE 1, compound C) was reacted with GdCl₃.6H₂O in H₂O, maintaining at pH 6.5 by addition of 1N meglumine. The

10 desired product was obtained.

The $^{1}H-NMR$, $^{13}C-NMR$, IR and MS spectra are consistent with the indicated structure.

EXAMPLE 13

- 15 $[[(3\beta,5\beta,7a,12a)-7,12-dihydroxy-3-[[4-[[2-[[4-[4,12$ bis(carboxy)-5,8,11-tris(carboxymethyl)-2-oxa-5,8,11triazadodecyl]phenyl]amino]-2-oxoethyl]amino]-1,4dioxobutyl]amino]-cholan-24-oate(6⁻)]gadolinate(3⁻)] hydrogen compound with 1-deoxy-1-(methylamino)-D-20 glucitol (1:3)
 - $(3\beta, 5\beta, 7a, 12a) 7, 12 dihydroxy 3 [[4 [[2 [[4 [4, 12 [4]]]]]]]]$ bis(carboxy)-5,8,11-tris(carboxymethyl)-2-oxa-5,-8,11-triazadodecyl]phenyl]amino]-2-oxoethyl]amino]-1,4-dioxobutyl]amino]-cholan-24-oic acid
- 25 solution of 4-[(1,1-dimethylethoxy)carbonyl]-5,8,11-tris[2-(1,1-dimethylethoxy)-2-oxoethyl]-1-[4-[[[[(1,1-dimethylethoxy)carbonyl]amino]acetyl]amino]phenyl]-2-oxa-5,8,11-triazatridecan-13-oic acid 1,1dimethylethyl ester (prepared according to the 30 procedure described in EXAMPLE 1) in anisole and CH₂Cl₂ was treated with trifluoroacetic acid. After 3 days the

77

reaction mixture was evaporated under reduced pressure, the residue was taken up into CH2Cl2 and evaporated again, repeating said procedure 2 more times. residue was then suspended in H_2O , neutralized at $O^{\circ}C$ 5 with 25% NH_4OH (w/w) and extracted with ethyl ether. The aqueous phase was evaporated under reduced pressure obtain a residue that was purified by chromatography. The resulting solid was dissolved in DMF and triethylamine and said solution was added with 10 the succinimido derivative of $(3\beta, 5\beta, 7\alpha, 12\alpha)-3-[[4-$ [(2,5-dioxo-1-pyrrolidinyl)oxy]-1,4-dioxobutyl]amino]-7,12-dihydroxy-cholan-24-oic acid methyl (prepared according to the procedure described Example 12). After 24h the reaction mixture was added 15 with a LiOH monohydrate aqueous solution, then the solvent was evaporated and the residue was dissolved in an Amberlite^R HCl and eluted through polystyrene resin, to obtain the desired product. The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent 20 with the indicated structure.

B) Title compound

According to the procedure described in EXAMPLE 1, compound A) was reacted with GdCl3.6H2O in H2O, maintaining at pH 6.5 by addition of 1N meglumine. The desired product was obtained.

The IR and MS spectra are consistent with the indicated structure.

EXAMPLE 14

25

 $[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[6-[[[bis[2-[bis(carboxymethyl)ami-$ 30 no]ethyl]amino]acetyl]amino]-1-oxohexyl]amino]-7,12-dihydroxy-cholan-24-oate(5⁻)]gadolinate(2⁻)] hydrogen

compound with 1-deoxy-1-(methylamino)-D-glucitol (1:2)

- A) N-[bis[2-[bis[[2-(1,1-dimethylethoxy)-2-oxoethyl]-amino]ethyl]glycine
- 1 g of glycine (marketed product) (0.0133 mol) was
 dissolved in 100 ml of H₂O/EtOH (25/75) and a solution
 of NaOH 1N (8.8 mL, 8.8 mmol) was added until pH = 10
 was reached. Then a solution of 10 g of N-(2bromoethyl)-N-[2-(1,1-dimethylethoxy)-2-oxoethyl]glycine 1,1 dimethylethyl ester (prepared according to
 Williams, M., A. et al., J. Org. Chem., 1993, 58, 1151)
 (0.0284 mmol) in 10 ml of 95% EtOH was added. After
 keeping the reaction at room temperature for 18 h., the
 mixture was evaporated to dryness. The residue was
 purified by flash chromatography obtaining 6 g of the

Yield 73%

desired product (0.010 mol).

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719) Eluent: 1 : 9 = MeOH : AcOH

Detection: 0.5% $KMnO_4$ in 0.1N NaOH $R_f = 0.20$

- 20 The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the structure.
 - B) (3β,5β,7α,12α)-3-[[[6-[(phenylmethoxy)carbonyl]amino]1-oxohexyl]amino]-7,12-dihydroxy-cholan-24oic acid methyl ester
- A solution of 3.46 g of N-Cbz-6-aminohexanoic acid (commercially available from Lancaster) (0.0130 mol) in 70 ml of THF and 1.8 ml of TEA (1.31 g, 0.0130 mol) was added, very quickly, with 1.77 g of isobutyl chloroformate (marketed product) (1.7 mL, 0.0130 mol) keeping the temperature at 0-3°C. After 15 min. 5 g of (3β,5β,7q,12a)-3-amino-7,12-dihydroxy-cholan-24-oic

79

acid methyl ester (prepared according to Example 5) (0.119 mol) in 30 ml of THF were added. After 30 min. from the end of dropping the temperature was kept at room temperature and the reaction mixture was filtered through a sintered glass filter and evaporated under reduced pressure. The solid was dissolved in $\mathrm{CH_2Cl_2}$ and washed with $\mathrm{H_2O}$ and brine. The phases were separated and the organic one was evaporated. The residue was dissolved in $\mathrm{CH_2Cl_2}$ and washed with a saturated solution $\mathrm{NaHCO_3}$ and $\mathrm{H_2O}$. The organic layers were combined, dried and evaporated under reduced pressure to give a solid, that was crystallized by AcOEt, obtaining the desired product (4.7 g, 0.0070 mol).

Yield: 60%

10

15 HPLC: 98.5 % (area %)

Stationary phase: Column E. Merck Lichrosorb Select

B; 5 μ m; 250 x 4 mm

Mobile phase: Gradient elution

A = 0.01 M aqueous solution in KH_2PO_4 and 0.017 M

in H_3PO_4

 $B = CH_3CN$

	mın	* A	* B
	0	95	5
	30	20	80
25	45	20	80

Flow rate:

 1 ml min^{-1}

Temperature:

45 °C

Detector (UV):

210 nm

Elemental analysis C H N

30 % calc.:

70.03 9.04 4.19

% found:

69.82 9.08 4.18

18 H₂O 0.24

WO 95/32741

80

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719)

Eluent: AcOEt : i-PrOH = 95 : 5 (v/v)

Detection: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100 :

2 : 1 (v/v/v) $R_{f} = 0.41$

- The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent 5 with the structure.
 - C) $(3\beta, 5\beta, 7\alpha, 12\alpha) 3 [(6-amino-1-oxohexyl)amino] 7, 12$ dihydroxy-cholan-24-oic acid methyl ester

4 g of compound B) (0.00598 mol) were dissolved in 50 ml of EtOH abs. and 800 mg of Pd/C were added. The 10 hydrogenation was performed at room temperature and atmospheric pressure. When the reaction has terminated, the mixture was filtered and evaporated to dryness. The residue was purified by flash chromatography obtaining

the desired product (2.7 g, 0.005 mol). 15

Yield: 84.4%

HPLC: 95 % (area %)

Stationary phase: Column E. Merck Lichrosorb Select

B; 5 µm; 250 x 4 mm

20 Mobile phase: Gradient elution

> A = aqueous solution 0.01 M in KH_2PO_4 and 0.017 M in H₃PO₄

 $B = CH_3CN$

	min	% A	% B
25	0	95	5
	30	20	80
	45	20	80

Flow rate: 1 ml min^{-1}

Temperature:

45 °C

Detector (UV): 210 nm

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719)

81

Eluent: MeOH : TEA = 95 : 5 (v/v)

Detection: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100 :

 $2 : 1 (v/v/v) R_f = 0.33$

5

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the structure.

D) (3β,5β,7α,12α)-3-[[6-[[[Bis[2-[bis(carboxymethyl)amino]ethyl]amino]acetyl]amino]-1-oxohexyl]amino]7,12-dihydroxy-cholan-24-oic acid

Equimolecular amounts of compound A) and compound C) were reacted at room temperature with 10 benzotriazol-1-yloxy-tris(dimethylamino)phosphonium hexafluorophosphate (BOP, marketed product) in DMF and in the presence of a N, N-diisopropylethylamine excess. When the reaction was over, the reaction mixture was 15 evaporated under vacuum and the residue taken up into EtOAc. The solution was washed with a NH₄Cl saturated solution and with H2O to neutral pH, then evaporated. The residue was hydrolysed first with 1M NaOH in MeOH/H2O then with CF3COOH in CH2Cl2, to give the 20 desired product that was purified and salted off by elution on an Amberlite $^{\rm R}$ XAD-16 resin using a MeOH/H $_2$ O gradient.

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the indicated structure.

25 E) Title compound

30

According to the procedure described in EXAMPLE 1, compound A) was reacted with ${\rm GdCl_3.6H_2O}$ in ${\rm H_2O}$, maintaining at pH 6.5 by addition of 1N meglumine. The desired product was obtained, that was salted off by elution with a MeOH/H₂O gradient on an Amberlyte XAD-16 resin.

82

The IR and MS spectra are consistent with the indicated structure.

In the same way the gadolinium complexes of the following ligands were prepared:

- 5 (3β,5β,7α,12α)-3-[[[[[bis[2-[bis(carboxymethyl)amino]-ethyl]amino]acetyl]amino]acetyl]amino]-7,12-dihydroxy-cholan-24-oic acid (COMPOUND 26);
 N⁶-[[bis[2-[bis(carboxymethyl)amino]ethyl]amino]ace-tyl]-N²-[(3α,5β,7α,12α)-3,7,12-trihydroxy-24-oxocholan-
- 10 24-yl]-L-lysine (COMPOUND 27);

EXAMPLE 15

- [[N⁶-[(4S)[4-[Bis[2-[bis(carboxymethyl)amino]ethyl]-amino]-4-carboxy]-1-oxobutyl]-N²-[(3 α ,5 β ,7 α ,12 α)-3,7,12-trihydroxy-24-oxocholan-24-yl]-L-lysinate(6⁻)]-
- 15 gadolinate(3-)] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:3)
 - A) N,N-Bis[2-[bis[2-(1,1-dimethylethoxy)-2-oxoethyl]amino]ethyl]-L-glutamic 1-(1,1-dimethylethyl)
 ester 5-(phenylmethyl)ester
- of 1-(1,1-dimethylethyl) ester 20 132.01 q (phenylmethyl)ester L-glutamic acid (prepared according to Helv. Chim. Acta, 199, 1864, 1958) (0.45 mol) were dissolved in 200 ml of H2O and 1L of EtOH and added to 320.2 g of N-(2-bromoethyl)-N-[2-(1,1-dimethylethoxy)-2-oxoethyl]glycine 1,1 dimethylethyl ester (prepared 25 according to Williams, M., A. et al., J. Org. Chem., 58, 1151. 1993) (0.909 mol) maintaining at pH 8 by addition of 10N NaOH. After 50h at 5°C, temperature was brought to 20°C for a further 80h and to 50°C for 5h. The mixture was adjusted to pH 7 with conc. HCl, 30

evaporated and extracted with hexane; the organic phase

83

was concentrated and the residue was purified by flash chromatography to obtain the desired product (75.2 g, 0.09 mol).

Yield: 20%

25

30

with the structure.

- 5 TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719) Eluent: hexane: AcOEt = 2: 1 (v/v) Detection: 0.5% KMnO₄ in 0.1N NaOH R_f = 0.76 The 1H-NMR, ¹³C-NMR, IR and MS spectra are consistent with the structure.
- 10 B) N,N-Bis[2-[bis[2-(1,1-dimethylethoxy)-2-oxoethyl]amino]ethyl]-L-glutamic acid 1-(1,1-dimethylethyl)
 ester
- 15.05 g of compound A) (18 mmol) dissolved in 100 ml of EtOH were added with 4 g of 5% wet Pd/C and the mixture was hydrogenated under a H₂ pressure of 111.36 kPa. When the reaction was over, the mixture was filtered, concentrated to a residue and purified by flash chromatography obtaining the desired product (11.01 g, 14.76 mmol).
- Yield: 82% $TLC: \ Silica \ gel \ plates \ 60 \ F_{254} \ (E. \ Merck \ art. \ 5719)$ Eluent: AcOEt $Detection: \ 0.5% \ KMnO_4 \ in \ 0.1N \ NaOH \qquad R_f = \ 0.85$ $The \ 1H-NMR, \ ^{13}C-NMR, \ IR \ and \ MS \ spectra \ are \ consistent$
 - C) $N^6-[(4S)[4-[Bis[2-[bis(carboxymethyl)amino]ethyl]-amino]-4-carboxy]-1-oxobutyl]-N^2-[(3a,5\beta,7a,12a)-3,7,12-trihydroxy-24-oxocholan-24-yl]-L-lysine Equimolecular amounts of compound B) and N^2-[(3a,5\beta,7a,12a)-3,7,12-trihydroxy-24-oxocholan-24-yl]-$

L-lysine methyl ester monohydrochloride (prepared as

10

30

8) were reacted at room described in EXAMPLE temperature with BOP in DMF and in the presence of a N, N-diisopropylethylamine excess. When the reaction was over, the reaction mixture was evaporated under vacuum and the residue taken up with EtOAc. The solution was washed with a $\mathrm{NH_4Cl}$ saturated solution and with $\mathrm{H_2O}$ to neutral pH, then evaporated. The residue was hydrolysed first with 1M NaOH in MeOH/H2O, then with ${\tt CF_3COOH}$ in ${\tt CH_2Cl_2}$ to give the desired product that was purified and salted off by elution on an AmberliteR XAD-16 resin using a MeOH/H₂O gradient.

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS were consistent with the structure.

D) Title compound

- 15 According to the procedure described in EXAMPLE 1, compound B) was reacted with $\mathrm{GdCl_3.6H_2O}$ in $\mathrm{H_2O}$, maintaining at pH 6.5 by addition of 1N meglumine. The resulting product was salted off by elution with a $\mathrm{MeOH/H_2O}$ gradient on an Amberlite^R XAD-16 resin.
- 20 The IR and MS spectra are consistent with the structure.

In the same way the gadolinium complex of the following ligand was prepared:

[3β(S),5β,7α,12α]-3-[4-carboxy-4-[bis[2-[bis(carboxyme-thyl)amino]-1-oxobutyl]amino]-7,12-dihydro-xy-cholan-24-oic acid (COMPOUND 29);

EXAMPLE 16

[[10-[2-[[2-[[(3\beta,5\beta,7\alpha,12\alpha)-7,12-dihydroxy-24-oxo-24-[(2-sulfoethyl)amino]cholan-3-yl]amino]-2-oxoethyl]amino]-2-oxoethyl]-1,4,7,10-tetraazacyclododecane-1,4,7triacetoate(4-)]gadolinate(1-)] hydrogen compound with

1-deoxy-1-(methylamino)-D-glucitol (1:1)

A) 1,4,7,10-Tetraazacyclododecane-1,4,7-triacetic acid tris(1,1-dimethylethyl)ester monohydrochloride

5 A stirred solution of 90 g of 10-formyl-1,4,7,10tetraazacyclododecane-triacetic acid tris(1,1-dimethylethyl)ester (prepared according to EP-A-292689) (0.166 mol) in 1L of anhydrous EtOH was added with 12.24 g of hydroxylamine hydrochloride (0.1826 mol) and refluxed 10 under argon atmosphere for 18 hours. At the end of this time, the reaction mixture was cooled and ethanol was removed under reduced pressure. To the resulting solid, $\mathrm{CH}_{2}\mathrm{Cl}_{2}$ was added and the suspension was transferred to a separatory funnel. After washing with water and brine, the organic phase was separated, dried and 15 concentrated under reduced pressure to obtain residue. The solid was recrystallized twice from a CH₂Cl₂ / hexane mixture and dried under in a vacuum oven at 35°C for 18 hours to obtain 57 g of the desired 20 product (0.103 mol).

Yield 62.3%

Elemental analysis C H N Cl % calc.: 55.21 9.25 9.84 7.49 % found: 55.40 9.43 9.84 7.48 H₂O 1.41

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719)

Eluent: 95 : 5 = MeOH : AcOH

Detection: 0.5% $KMnO_4$ in 0.1N NaOH R_f = 0.67

The 1H -NMR, ^{13}C -NMR, IR and MS spectra are consistent with the structure.

30 B) (3β,5β,7α,12α)-3-[[[(1,1-dimethylethoxy)carbo-nyl]amino]acetyl]amino]-7,12-dihydroxy-cholan-24-

oic acid methyl ester

To a solution of N-(t-butoxycarbonyl)glycine (marketed product) (14.7 g; 84.0 mmol) and Et_3N (8.50 g; 11.6 ml; 84.0 mmol) in THF (400 ml), at 0°C and under nitrogen, was added dropwise isobutyl chloroformate (11.5 g; 10.9 ml; 84.0 mmol). After 15 min a solution $(3\beta,5\beta,7\alpha,12\alpha)-3-amino-7,12-dihydroxy-cholan-24-oic$ acid methyl ester (prepared according to Example 5) (29.5 g; 70.0 mmol) in THF (100 ml) was added dropwise to the reaction mixture at 0°C. After 20 min the 10 reaction mixture was allowed to rise room temperature and stirred overnight. The suspension was filtered sintered funnel and the filtrate through a evaporated off under reduced pressure to give a residue 15 that was dissolved in Et₂O and washed with a saturated. aqueous solution of NaHCO3 and H2O. The organic phase was separated, dried and then evaporated under reduced The solid residue was purified by flash chromatography to give the desired product as a white 20 solid (25.3 g; 43.7 mmol).

Yield: 62% m.p.: 110-114°C

K.F.: 0.75 %

HPLC: 97 % (area %)

Stationary phase: Column E. Merck Lichrosorb Select

25 B; 5 µm; 250 x 4 mm

Mobile phase: Gradient elution

A = aqueous solution 0.01 M in KH_2PO_4 and 0.017 M in H_3PO_4

 $B = CH_3CN$

87

min	% A	% B
0	95	5
30 .	20	80
45	20	80

Flow rate:

 1 ml min^{-1}

Temperature:

45 °C

Detector (UV):

210 nm

Elemental analysis C

66.40 9.40 N 4.84

% calc.:

H

10 % found:

64.97

9.07

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719) Eluent: EtOAc : i-PrOH = 9:1 (v/v)

Detection: AcOH : Conc. H_2SO_4 : p-anisaldehyde =

100:2:1 (v/v/v)

 $R_{f} = 0.43$

- The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent 15 with the structure.
 - C) $(3\beta, 5\beta, 7\alpha, 12\alpha)-3-[[[[(1,1-dimethylethoxy)carbo$ nyl]amino]acetyl]amino]-7,12-dihydroxy-cholan-24oic acid
- 20 To a solution of compound B) (24.5 g; 41.1 mmol) in MeOH/H2O (2:1, v/v; 160 ml) at room temperature, was added dropwise 1N NaOH (49.7 ml; 49.7 mmol) in 2 hours. After 48 h the reaction mixture was filtered through a sintered glass filter and evaporated under reduced pressure. The residue was treated with 0.5 N HCl/EtOAc 25 (1:2, 240 ml) and pH of the resulting mixture was adjusted to 2 with 2N HCl (10 ml) under vigorous stirring. After separation the aqueous phase saturated with NaCl and extracted with EtOAc. organic layers were combined, dried and evaporated 30 under reduced pressure obtaining the desired product

88

```
(22.2 g, 39.3 mmol).
```

Yield: 96% m.p.: 150-155°C

K.F.: 0.75 %

Acidic titre (0.1 N NaOH): 95 %

5 HPLC: 96 % (area %)

Stationary phase: Column E. Merck Lichrosorb Select

B; 5 µm; 250 x 4 mm

Mobile phase: Gradient elution

A = aqueous solution 0.01 M in KH_2PO_4 and 0.017 M

in H₃PO₄

 $B = CH_3CN$

min	% A	% E
0	95	5
30	20	80
45 ·	20	80

Flow rate:

15

 1 ml min^{-1}

Temperature: 45 °C

Detector (UV): 210 nm

Elemental analysis ${\tt C}$ ${\tt H}$ ${\tt N}$

20 % calc.: 65.92 9.28 4.96 % found: 65.41 9.98 4.60

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719)

Eluent: EtOAc : i-PrOH : AcOH = 90 : 15 : 1 (v/v)

Detection: AcOH : Conc. H₂SO₄ : p-anisaldehyde =

25 100:2:1 (v/v/v) $R_f = 0.47$

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the structure.

- D) $2-[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[[(1,1-dimethylethoxy)car-bonyl]amino]acetyl]amino]-7,12-dihydroxy-24-oxo-$
- 30 cholan-24-yl]amino]ethanesulfonic acid

Taurine (2-aminoethanesulfonic acid) (marketed

product) (5.40 g; 43.1 mmol) and Et_3N (5.16 g; 7.07 ml; 51.0 mmol) were added to a solution of compound C) (22.1 g; 39.1 mmol) and (2-ethoxy-1-ethoxycarbonyl-1,2dihydroquinoline, EEDQ) (marketed product) (12.6 g; 5 (100 ml) under in DMF nitrogen. resulting suspension was heated at 90°C for 70 min. obtaining a clear solution which was then cooled to 25°C. After 30 min the reaction mixture was poured slowly into cold Et₂O (0°C, 900 ml): a resinous product 10 was formed. The suspension was kept at 4°C overnight. The mixture was decanted and the resinous substance was washed with $\mathrm{Et}_2\mathrm{O}$, treated with $\mathrm{CH}_2\mathrm{Cl}_2$ and filtered to remove unreacted taurine. The filtrate was poured into cold $\mathrm{Et}_2\mathrm{O}$ (0°C) the precipitate was filtered through a sintered glass filter and immediately dissolved in 0.4 15 N NaOH in MeOH (100 ml). After diluting the solution with Et₂O, the suspension was kept at 4°C for several hours and then filtered through a sintered glass filter. The solid was washed thoroughly with $\mathrm{Et}_2\mathrm{O}$ and 20 dried under reduced pressure to give the desired product (24.7 g; 35.6 mmol).

Yield: 91% m.p.: 150-155°C

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719)

Eluent: $CHCl_3$: MeOH : AcOH = 90 : 30 : 4 (v/v/v)

Detection: AcOH : Conc. $\rm H_2SO_4$: p-anisaldehyde = 100 : 2 : 1 (v/v/v) $\rm R_f$ = 0.16 The $^1\rm H-NMR$, $^{13}\rm C-NMR$, IR and MS spectra are consistent with the structure.

E) 2-[[(3β,5β,7α,12α)-3-[(aminoacetyl)amino]-7,12-di 30 hydroxy-24-oxocholan-24-yl]amino] ethanesulfonic
 acid sodium salt

PCT/EP95/01958

90

22.4 g of compound D) (32.3 mmol) were suspended in 1M methanolic HCl (160 mmol, 160 ml) at room temperature. During the reaction time, the suspension became thicker and after 1 day the reaction mixture was filtered through a sintered glass filter. The collected solid was washed thoroughly with $\rm Et_2O/MeOH$ (1:1 v/v) and dried under reduced pressure obtaining the desired product (13.0 g; 20.6 mmol).

Yield: 64% m.p.: 200°C

10 HPLC: 94% (area %)

WO 95/32741

Stationary phase: Column E. Merck Lichrosorb Select B; 5 μm ; 250 x 4 mm

Mobile phase: Gradient elution

A = aqueous solution 0.01 M in KH_2PO_4 and 0.017 M

in H_3PO_4

20

 $B = CH_3CN$

min	% A	% B
0	95	5
30	20	80
45	20	80

Flow rate: 1 ml min⁻¹

Temperature: 45 °C

Detector (UV): 210 nm

TLC: Silica gel plates 60 F_{254} (E. Merck art. 5719)

25 Eluent: MeOH : AcOH = 95 : 5 (v/v)

Detection: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100 :

2 : 1 (v/v/v) $R_f = 0.67$

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the structure.

30 F) 10-[2-[[2-[[(3β,5β,7α,12α)-7,12-dihydroxy-24-oxo-24-[(2-sulfoethyl)amino]cholan-3-yl]amino]-2-oxo-

91

ethyl]amino]-2-oxoethyl]-1,4,7,10-tetraazacyclododecane-1,4,7-triacetic acid

A solution of compound A) and triethylamine in DMF 5°C drop by drop with was added isobutyl chloroformate and subsequently with a solution compound E) in DMF. When the reaction was over solvent was evaporated under vacuum, the residue was dissolved with CH2Cl2 and trifluoroacetic acid was dropped therein at 0°C. When the addition was completed the mixture was left to react at room temperature. When the reaction was over the reaction mixture was evaporated under reduced pressure. The residue was taken up with $\mathrm{CH}_{2}\mathrm{Cl}_{2}$ and evaporated again repeating such a procedure 2 more times. The residue was purified and salted off by elution on an Amberlite R XAD-16 resin using a $\ensuremath{\text{MeOH/H}_2\text{O}}$ gradient to obtain the desired product. The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent

with the structure.

G) Title compound

5

10

15

20 According to the procedure described in EXAMPLE 1, compound F) was reacted with GdCl3.6H2O in H2O, maintaining at pH 6.5 by addition of 1N meglumine. The desired product was obtained that was salted off by elution with a $MeOH/H_2O$ gradient on an Amberlyte XAD-16 25 resin.

The ĪR and MS spectra are consistent with the structure.

In the same way, the gadolinium complexes of the following ligands were prepared:

30 $(3\beta, 5\beta, 7\alpha, 12\alpha)-3-[[[[[4,7,10-tris(Carboxymethyl)-$ 1,4,7,10-tetraazacyclododecyl]acetyl]amino]acetyl]amino]-7,12-dihydroxy-cholan-24-oic acid (COMPOUND 31); $N^2-[(3\alpha,5\beta,7\alpha,12\alpha)-3,7,12-\text{trihydroxy-}24-\text{oxocholan-}24-\text{yl}]-N^6-[[4,7,10-\text{tris}(\text{carboxymethyl})-1,4,7,10-\text{tetraaza-cyclododecyl}]acetyl]-L-lysine (COMPOUND 32);$

5 (3β,5β,7α,12α)-3-[[6-[[[4,7,10-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecyl]acetyl]amino]-1-oxo-hexyl]amino]-7,12-dihydroxy-cholan-24-oic acid (COM-POUND 33).

EXAMPLE 17

- [[(3α,5β,7α,12α)-3-[[3-[4,7,10-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecyl]-2-hydroxypropyl]oxy]-7,12dihydroxy-cholan-24-oate(4⁻)]gadolinate(1⁻)] hydrogen
 compound with 1-deoxy-1-(methylamino)-D-glucitol (1:1)
- compound with 1-deoxy-1-(methylamino)-D-glucitol (1:1) $(3a,5\beta,7a,12a)-3-(2,3-epoxypropyl)oxy-7,12-dihy-$ A) 15 droxy-cholan-24-oic acid 1,1-dimethylethyl ester A mixture of 50% NaOH (10 ml), epichlorohydrin (6 ml) and tetrabutylammonium hydrogen sulfate (0.3 g) kept at 0°C was added drop by drop with a solution of $(3\alpha, 5\beta, 7\alpha, 12\alpha)-3, 7, 12-trihydroxy-cholan-24-oic$ acid 1,1-dimethylethyl ester (prepared according to the 20 procedure described by R. P. Bonar-Law et al., J. Chem. Soc. Perkin Trans. I, 1990, 2245) (0.0045 mol) CH₂Cl₂ (10 ml). When the addition was completed the mixture was left to react at room temperature by 24 h.
- The organic phase was separated, washed with $\rm H_2O$ to neutral, dried over $\rm Na_2SO_4$ and evaporated. The residue was purified by flash chromatography to obtain 0.86 g of desired product (0.0017 mol).

Yield: 37%

30 TLC: Carrier: silica gel plates 60 F_{254} Merck Eluent: n-hexane: AcOEt = 1:1 (v/v)

93

Detector: AcOH : Conc. H_2SO_4 : p-anisaldehyde = 100:2:1 R_f = 0.3

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the structure.

B) (3α,5β,7α,12α)-3-[[3-[4,7,10-tris[2-(1,1-dimethyl-ethoxy)-2-oxoethyl]-1,4,7,10-tetraazacyclodode-cyl]-2-hydroxypropyl]oxy]-7,12-dihydroxy-cholan-24-oic acid (1,1-dimethylethyl)ester

A solution containing compound A) (0.001 mol),

10 1,4,7,10-tetraazacyclododecane-1,4,7-triacetic acid

tris(1,1-dimethylethyl)ester monohydrochloride (prepared according to Example 16) (0.001 mol) and

triethylamine (1.5 ml) in EtOH (30 ml) was refluxed for

4 h. The reaction mixture was evaporated and the

residue was purified by flash chromatography to obtain

Yield: 27%

30

TLC: Carrier: silica gel plates 60 F_{254} Merck Eluent: CH_2Cl_2 : MeOH = 9 : 1 (v/v)

0.3 g of desired product (0.0003 mol).

Detector: AcOH: Conc. H_2SO_4 : p-anisaldehyde = 100:2:1 R_f = 0.34

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the structure.

C) (3α,5β,7α,12α)-3-[[3-[4,7,10-tris(carboxymethyl) 1,4,7,10-tetraazacyclododecyl]-2-hydroxypropyl] oxy]-7,12-dihydroxy-cholan-24-oic acid

A solution of compound B) in $\mathrm{CH_2Cl_2}$ at 0°C was added drop by drop with trifluoroacetic acid. When the addition was completed the mixture was left to react at room temperature. When the reaction was over the reaction mixture was evaporated under reduced pressure.

The residue was taken up with $\mathrm{CH_2Cl_2}$ and evaporated again repeating such a procedure 2 more times. The residue was purified and salted off by elution on an Amberlite^R XAD-16 resin using a MeOH/H₂O gradient to obtain the desired product.

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the structure.

D) Title compound

According to the procedure described in EXAMPLE 1, compound C) was reacted with $\mathrm{GdCl}_3.6\mathrm{H}_2\mathrm{O}$ in $\mathrm{H}_2\mathrm{O}$, maintaining at pH 6.5 by addition of 1N meglumine. The resulting product was salted off by elution with a MeOH/H $_2\mathrm{O}$ gradient on an Amberlite R XAD-16 resin.

The IR and MS spectra are consistent with the structure.

EXAMPLE 18

15

[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[5-[4,7,10-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecyl]-4-hydroxy-1-oxopentyl]-amino]-7,12-dihydroxy-cholan-24-oate(4-)]gadolinate-

- 20 (1-)] hydrogen compound with 1-deoxy-1-(methylamino)-D-glucitol (1:1)
 - A) (3β,5β,7α,12α)-3-(1-oxopent-4-enyl)amino-7,12-dihydroxy-cholan-24-oic acid methyl ester

A solution of 4-pentenoic acid (marketed product)

and triethylamine in THF was added drop by drop, under
nitrogen and at 5°C, with isobutyl chloroformate and
subsequently with a solution of (3β,5β,7α,12α)-3-amino7,12-dihydroxy-cholan-24-oic acid methyl ester (prepared according to the procedure described in EXAMPLE 5)

in THF. When the reaction was over, solvent was
evaporated and the residue was taken up with H₂O and

95

AcoEt. The organic phase was separated, washed with $\rm H_{2}O$, dried over $\rm Na_{2}SO_{4}$ and evaporated under reduced pressure. The residue was purified by flash chromatography to obtain the desired product.

- 5 The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra are consistent with the structure.
 - B) (3β,5β,7α,12α)-3-(4,5-epoxy-1-oxopentyl)amino-7,12-dihydroxy-cholan-24-oic acid methyl ester

A solution of magnesium monoperphthalate in H2O was dropped into a solution of compound A) in CHCl3 10 containing methyltrioctylammonium chloride and kept at 50°C. The pH of the reaction mixture was maintained from 4.5 to 5 by addition of 5% NaOH. When the reaction was over the organic phase was separated and the aqueous phase was extracted with CHCl3. The organic 15 phases were combined, washed with H20, dried over Na2SO4 and evaporated after checking for the absence of peroxides. The residue was purified by flash chromatography to obtain the desired product.

The $^{1}\text{H-NMR}$, $^{13}\text{C-NMR}$, IR and MS spectra were consistent with the structure.

25

30

C) (3β,5β,7α,12α)-3-[[5-[4,7,10-tris[2-(1,1-dimethyl-ethoxy)-2-oxoethyl]-1,4,7,10-tetraazacyclodode-cyl]-4-hydroxy-1-oxopentyl]amino]-7,12-dihydroxy-cholan-24-oic acid

A solution containing B), 1,4,7,10-tetraazacyclo-dodecane-1,4,7-triacetic acid tris(1,1-dimethylethyl)-ester monohydrochloride (prepared according to Example 16) and triethylamine in EtOH was refluxed for 4 h. The reaction mixture was evaporated and the residue was purified by flash chromatography to obtain the desired

PCT/EP95/01958 WO 95/32741

96

product.

5

10

15

25

30

The 1H-NMR, 13C-NMR, IR and MS spectra were consistent with the structure.

 $(3\beta, 5\beta, 7\alpha, 12\alpha)-3-[[5-[4,7,10-tris(carboxymethyl)-$ 1,4,7,10-tetraazacyclododecyl]-4-hydroxy-1-oxopentyl]amino]-7,12-dihydroxy-cholan-24-oic acid

A solution of compound C) in CH2Cl2 at 0°C was added drop by drop with trifluoroacetic acid. When the addition was completed the mixture was left to react at room temperature. When the reaction was over the reaction mixture was evaporated under reduced pressure. The residue was taken up with CH₂Cl₂ and evaporated again, repeating such a procedure 2 more times. The residue was purified and salted off by elution on an Amberlite^R XAD-16 resin using a MeOH/H₂O gradient to obtain the desired product.

The 1H-NMR, 13C-NMR, IR and MS spectra were consistent with the structure.

Title compound E)

According to the procedure described in EXAMPLE 1, 20 compound D) was reacted with GdCl₃.6H₂O in H₂O, maintaining at pH 6.5 by addition of 1N meglumine. The resulting desired product was salted off by elution with a $MeOH/H_2O$ gradient on an Amberlyte XAD-16 resin.

and MS spectra were consistent with the IR structure.

EXAMPLE 19

The relaxivities r1 and r2 $(mM^{-1}.s^{-1})$ of the 4 $carboxy-5,8,11-tris(carboxymethyl)-1-[4-[[[(3a,5\beta,7a,-$ 12a)-3,7,12-trihydroxy-24-oxocholan-24-yl]amino]acetyl]amino]phenyl]-2-oxa-5,8,11-triazatridecan-13-oic

97

acid gadolinium complex (Compound 1F) were evaluated in SERONORM-HUMANTM serum (NYCOMED), in a magnetic filed of a 20 MHz frequency, at a temperature of 39°C, (MINISPEC PC-120 device), using the following sequences: Inversion Recovery; CPMG; and compared with those of Gd-DTPA/Dimeg (Magnevist^R), Gd-DOTA/meg (Dotarem^R), Gd-BOPTA/Dimeg and GdCl $_3$ [percent ratios being calculated with respect to GdCl $_3$] obtained under the same experimental conditions. The results are reported in Table 1.

Table 1

 $A = \frac{r_1(Gd-compound)}{r_1(GdCl_3)}$ $B = \frac{r_2(Gd-compound)}{r_2(GdCl_3)}$

5

Compounds	$r_1 (mM^{-1}.s^{-1})$	A.100 -	r ₂ (mM ⁻¹ .s ⁻¹)	B.100
Compound 1F	19.23	183.8	22.02	182.1
Compound 4F	12.02	114.9	13.74	113.6
Magnevist [©]	4.96	47.42	5.43	44.91
Dotarem [®]	4.34	41.49	5.02	41.52
Gd-BOPTA	9.31	89.00	11.19	92.55
GdCl ₃	10.46	100	12.09	100

WO 95/32741

98

CLAIMS

1. Compounds of general formula (I)

A-L-B (I),

5 wherein

A is the residue of a bile acid, wherein by bile acid the group of the bile acids obtainable by bioconversion from cholesterol is meant, particularly the acids: cholic, deoxycholic, chenodeoxycholic, ursodeoxycholic, lithocholic, and the derivatives thereof, including those with taurine and glycine;

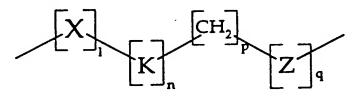
L is a ligand between one of the C-3, C-7, C-12 or C-24 positions of the residue of the bile acid and B, corresponding to a group of formula (II)



in which

20 m is an integer varying from 1 to 10, wherein for values above 1, Y can have different meanings,

Y corresponds to the following succession of groups,



25

10

15

n, 1 and q can be 0 or 1,

p can vary from 0 to 10,

X is an O atom, a S atom, or a -NR group,

30 in which

R is a H atom, or a (C_1-C_5) alkyl group,

99

K is benzene ring, substituted or not, or a $-\text{CHR}_1$ group,

wherein

5

20

 R_1 is an hydrogen atom, or a -COOH group, or a -SO $_3$ H group,

- Z is an O atom or a S atom, or one of the -CO- or -CS- groups,
- B is the residue of a chelating agent of the bitrivalent metal ions having an atomic number varying from 20 to 31, 39, 42, 43, 44, 49, or from 57 to 83, wherein said residue can in its turn be conjugated or not, by a second chain L of formula (II), to another residue A as defined above,
- with the proviso that at least one from 1, n, q, p is

 different from 0 and in case X and Z are both 0 or S

 atoms, q or n is equal to 1,
 - as well as complex chelates of said compounds of formula (I) with ions of metal elements having atomic number ranging from 20 to 31, 39, from 42 to 44, 49 and from 57 to 83 and the salts thereof with physiologically acceptable organic bases selected from primary, secondary, tertiary amines or basic amino acids, or with inorganic bases whose cations are sodium, potassium, magnesium, calcium or mixtures thereof, or with physiologically acceptable anions of
- thereof, or with physiologically acceptable anions of organic acids selected from acetate, succinate, citrate, fumarate, maleate, oxalate, or with anions of inorganic acids selected from halohydric acid ions.
- Compounds according to claim 1, wherein A is
 selected from one of the following bile acids: cholic, deoxycholic, chenodeoxycholic, ursodeoxycholic, lito-

100

cholic and drivatives thereof, including those conjugated with taurine and glycine.

- Compounds according to claim 1, wherein B selected from the following polyaminocarboxylic acids and ester or amide derivatives thereof: EDTA; DTPA; EOB-DTPA; BOPTA; DTPA-BMA; DOTA; DOTMA; DO3A; HPDO3A; MCTA; or B is selected from the following acids: DPDP; following EDTP; or В is selected from the polyaminophosphonic acids and derivatives thereof or polyaminophosphinic acids and derivatives thereof: 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetrakis[methylene(methylphosphinic)] acid and 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetrakis[methylene(methylphosphonic)] acid; or B is selected from macrocyclic chelants selected from texafirines, porphyrins and phthalocyanines.
- 4. Compounds according to claim 1, wherein the spacing chain L has the following general formula (III),

20

5

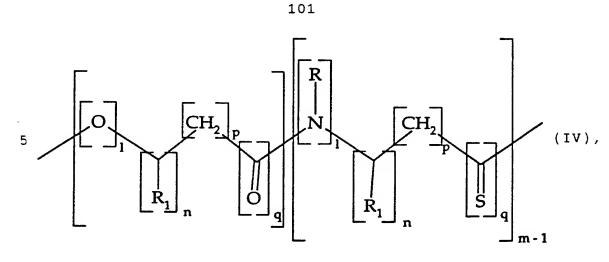
10

15

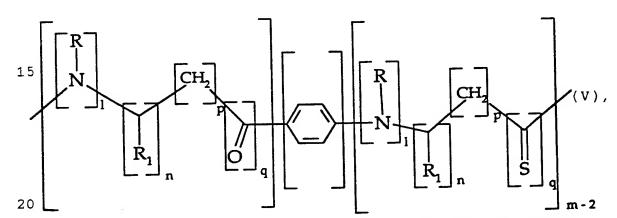
$$\begin{bmatrix} R \\ N \end{bmatrix}_1 \begin{bmatrix} CH_2 \\ P \end{bmatrix}_{p}$$
 (III),

25

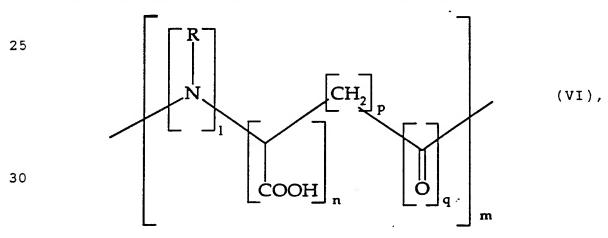
wherein R, R_1 , l, m, n, p and q are as defined above. 5. Compounds according to claim 1, wherein the spacing chain L has the following general formula (IV),



wherein R, R_1 , l, m, n, p and q are as defined above. 10 Compounds according to claim 1, wherein the spacing chain L has the following general formula (V),



wherein R, R_1 , l, m, n, p and q are as defined above. wherein the Compounds according to claim 1, 7. spacing chain L has the following general formula (VI),



102

wherein R, l, m, n, p and q are as defined above.

- 8. Compounds according to claim 1, wherein bi- or trivalent metal ion complexed by the chelant residue B is selected from $Fe(^{2+})$, $Fe(^{3+})$, $Gd(^{3+})$, $Eu(^{3+})$, $Dv(^{3+})$, $Ia(^{3+})$, $Vb(^{3+})$ and $Mn(^{2+})$, or is a radioisotope
- ${
 m Dy(^{3+}),\ La(^{3+}),\ Yb(^{3+})\ and\ Mn(^{2+}),\ or\ is\ a\ radioisotope}$ selected from ${
 m ^{51}Cr,\ ^{67}Ga,\ ^{68}Ga,\ ^{111}In,\ ^{99m}Tc,\ ^{140}La,\ 175_{Yb},\ 153_{Sm},\ 166_{Ho},\ ^{90}_{Y},\ 149_{Pm},\ ^{177}Lu,\ ^{47}Sc,\ ^{142}_{Pr},\ 159_{Gd},\ ^{212}Bi.}$
- 9. Compounds according to claim 1, wherein the physiolagically acceptable salifying organic base is selected from ethanolamine, diethanolamine, morpholine, glucamine, N,N-dimethylglucamine, N-methylglucamine, lysine, arginine, ornithine.
- 10. Compounds according to claim 1, wherein physiolagically acceptable salifying inorganic acid anion is a halohydric acid ion selected from chlorides, bromides and iodides.

- 11. Compounds according to claim 1 wherein A is a residue of cholic acid or of a derivative thereof and B is a residue of DTPA or of a derivative thereof.
- 12. Compounds according to claim 1 wherein A is a residue of cholic acid or of a derivative thereof and B is a residue of BOPTA or of a derivative thereof.
- 13. Compounds according to claim 1 wherein A is a residue of cholic acid or of a derivative thereof and B is a residue of DOTA or of a derivative thereof.
 - 14. A compound according to claims 1 to 13, wherein the derivative of formula (I) is selected from the following:
- 30 Comp. 1 [[4-carboxy-5,8,11-tris(carboxymethyl)-1-[4- $[[[(3\alpha,5\beta,7\alpha,12\alpha)-3,7,12-trihydroxy-24-oxo-$

		103
		cholan-24-yl]amino]acetyl]amino]phenyl]-2-
		oxa-5,8,11-triazatridecan-13-oic acid;
	Comp. 2	<pre>[[4-carboxy-5,8,11-tris(carboxymethyl)-1-[4-</pre>
		[[(3a,5β,7a,12a)-3,7,12-trihydroxy-24-oxo-
5		cholan-24-yl]amino]phenyl]-2-oxa-5,8,11-tria-
		zatridecan-13-oic acid;
	Comp. 3	[[3,6,9-tris(carboxymethyl)-10-(phenylmetho-
		$xy)$ methyl-11-oxo-14-[[(3a,5 β ,7a,12a)-3,7,12-
		trihydroxy-24-oxocholan-24-y1]amino]-3,6,9,-
10		12-tetraazatetradecanoic acid;
	Comp. 4	[[10-[2-oxo-2-[[3-[[2-[[(3a,57a,12a)-3,7,-
		12-trihydroxy-24-oxocholan-24-yl]amino]-
		ethyl]amino]propyl]amino]ethyl]-1,4,7,10-
		tetraazacyclododecan-1,4,7-triacetic acid;
15	Comp. 5	[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[13-carboxy-6,9,12-tris-
		(carboxymethyl)-1,4-dioxo-5-[(phenylmethoxy)-
		methyl]-3,6,9,12-tetraazatridecyl]amino]-7,-
		12-dihydroxy-cholan-24-oic acid;
	Comp. 6	$[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[17-carboxy-10,13,16-$
20		tris(carboxymethyl)-8-oxo-9-[(phenylmetho-
		xy)methyl]-3,7,10,13,16-pentaazaheptadecyl]-
		oxy]-7,12-dihydroxy-cholan-24-oic acid;
	Comp. 7	$[[(3\beta,5\beta,7\alpha,12\alpha)-7,12-dihydroxy-3-[2-[[[4-$
		<pre>[4,12-bis(carboxy)-5,8,11-tris(carboxyme-</pre>
25		thyl)-2-oxa-5,8,11-triazadodecyl]phenyl]ami-
		no]thioxomethyl]amino]ethoxy]-cholan-24-oic
		acid;
	Comp. 8	$(3\beta, 5\beta, 7a, 12a) - 7, 12 - dihydroxy - 3 - [[[[3 - [[[4, -$
		7,10-tris(carboxymethyl)-1,4,7,10-tetraazacy-
30		<pre>clodec-1-yl]acetyl]amino]propyl]amino]ace-</pre>
		tyl]amino]-cholan-24-oic acid;

	Comp.	9	[[3,6,9-tris(carboxymethyl)-10-[(phenylmetho-
			$xy)methyl]-11-oxo-17-[[(3a,5\beta,7a,12a)-3,7,12-$
			trihydroxy-24-oxocholan-24-yl]amino]-3,6,9,-
			12-tetraazaoctadecanedioic acid;
5	Comp.	10	[[(3β,5β,7α,12α)-3-[[13-carboxy-6,9,12-tris-
			(carboxymethyl)-1,4-dioxo-3,6,9,12-tetraaza-
			tridecyl]amino]-7,12-dihydroxy-cholan-24-oic
			acid;
	Comp.	11	[(3\beta,5\beta,7\alpha,12\alpha)-(3'\beta,5'\beta,7'\alpha,12'\alpha)-3,3'-[[6,-
10			9,12-tris(carboxymethyl)-1,4,14,17-tetraoxo-
			3,6,9,12,15-pentaazaheptadecan-1,17-diyl]bi-
			simino]bis[7,12-dihydroxycholan-24-oic acid;
	Comp.	12	[[[3 β (S),5 β ,7a,12a]-7,12-dihydroxy-3-[[4-
			[[[[5-[bis[2-[bis(carboxymethyl)amino]ethyl]-
15			amino]-5-carboxypentyl]amino]thioxomethyl]-
			amino]benzoyl]amino]-cholan-24-oic acid;
	Comp.	13	$[[[3\beta(S),5\beta,7\alpha,12\alpha]-7,12-dihydroxy-3-[[4-[[5-$
			[bis[2-[bis(carboxymethyl)amino]ethyl]amino]-
			5-carboxypentyl]amino]-1,4-dioxobutyl]amino]-
20			cholan-24-oic acid;
	Comp.	14	[[(3β,5β,7α,12α)-7,12-dihydroxy-3-[[4-[[2-
			[[4-[4,12-bis(carboxy)-5,8,11-tris(carboxyme-
			thyl)-2-oxa-5,8,11-triazadodecyl]phenyl]ami-
			no]-2-oxoethyl]amino]-1,4-dioxobutyl]amino]-
25			cholan-24-oic acid;
	Comp.	15	$3,6,9$ -tris(carboxymethyl)- 14 -[[($3\beta,5\beta,7\alpha$,-
			12a)-7,12-dihydroxy-24-oxo-24-[(2-sulfo-
			ethyl)amino]-cholan-3-yl]amino]-11,14-dioxo-
			3,6,9,12-tetraazatetradecanoic acid;
30	Comp.	16	$N-[(3\beta,5\beta,7\alpha,12\alpha)-3-[[13-carboxy-6,9,12-tris-$
			(carboxymethyl)-1,4-dioxo-5-[(phenylmethoxy)-

			100
			methyl]-3,6,9,12-tetraazatridecyl]amino]-7,-
			12-dihydroxy-24-oxocholan-24-yl]glycine
	Comp.	17	(3β,5β,7α)-3-[[13-carboxy-6,9,12-tris(carbo-
			xymethyl)-1,4-dioxo-5-[(phenylmethoxy)me-
5			thyl]-3,6,9,12-tetraazatridecyl]amino]-7-hy-
			droxy-cholan-24-oic acid;
	Comp.	18	(3β,5β,12α)-3-[[13-carboxy-6,9,12-tris(carbo-
			xymethyl)-1,4-dioxo-5-[(phenylmethoxy)me-
			thyl]-3,6,9,12-tetraazatridecyl]amino]-12-hy-
10			droxy-cholan-24-oic acid;
	Comp.	19	(3β,5β)-3-[[13-carboxy-6,9,12-tris(carboxyme-
			thyl)-1,4-dioxo-5-[(phenylmethoxy)methyl]-3,-
			6,9,12-tetraazatridecyl]amino]-cholan-24-oic
			acid;
15	Comp.	20	(3β,5β,7α,12α)-3-[[17-carboxy-10,13,16-tris-
			(carboxymethyl)-1,8-dioxo-9-[(phenylmethoxy)-
			methyl]-7,10,13,16-tetraazaheptadecyl]amino]-
			7,12-dihydroxy-cholan-24-oic acid;
	Comp.	21	(3β,5β,7α,12a)-7,12-dihydroxy-3-[[3-[[[4,-
20			7,10-tris(carboxymethyl)-1,4,7,10-tetraazacy-
			<pre>clododec-1-yl]acetyl]amino]propyl]amino]-cho-</pre>
			lan-24-oic acid;
	Comp.	22	$3,6,9$ -tris(carboxymethyl)- 14 -[[(3 β ,5 β ,7 α ,-
			12a)-7,12-dihydroxy-24-oxo-24-[(2-sulfo-
25			ethyl)amino]-cholan-3-yl]amino]-11,14-dioxo-
			3,6,9,12-tetraazatetradecanoic acid;
	Comp.	23	[(3\beta,5\beta,7\alpha,12\alpha)-3-[[17-carboxy-10,13,16-tris-
			(carboxymethyl)-1,8-dioxo-7,10,13,16-tetra-
			azaheptadecyl]amino]-7,12-dihydroxy-cholan-
30			24-oic acid;
	Comp.	24	(17S)-3,6,9-tris(carboxymethyl)-11-oxo-17-

106

 $[[(33,5\beta,7a,12a)-3,7,12-trihydroxy-24-oxocho$ lan-24-yl]amino]-3,6,9,12-tetraazaoctadecanedioic acid; Comp. 25 $[[(3\beta,5\beta,7\alpha,12\alpha)-3-[[6-[[[bis[2-[bis(carboxy-$ 5 methyl)amino]ethyl]amino]acetyl]amino]-1-oxohexyl]amino]-7,12-dihydroxy-cholan-24-oic acid; Comp. 26 $(3\beta, 5\beta, 7\alpha, 12\alpha) - 3 - [[[[bis[2-[bis(carboxyme$ thyl)amino]ethyl]amino]acetyl]amino]acetyl]amino]-7,12-dihydroxy-cholan-24-oic acid; 10 N^{6} -[[bis[2-[bis(carboxymethyl)amino]ethyl]-Comp. 27 amino]acetyl]- N^2 -[(3a,5 β ,7a,12a)-3,7,12-trihydroxy-24-oxocholan-24-yl]-L-lysine $[[N^6-[(4S)[4-[bis[2-[bis(carboxymethyl)ami-$ Comp. 28 no]ethyl]amino]-4-carboxy]-1-oxobutyl]- N^2 -15 $[(3a,5\beta,7a,12a)-3,7,12-trihydroxy-24-oxocho$ lan-24-yl]-L-lysine $[3\beta(S), 5\beta, 7\alpha, 12\alpha]-3-[4-carboxy-4-[bis[2-[bis-$ Comp. 29 (carboxymethyl)amino]ethyl]amino]-1-oxobu-20 tyl]amino]-7,12-dihydroxy-cholan-24-oic acid; $[[10-[2-[[2-[[(3\beta,5\beta,7\alpha,12\alpha)-7,12-dihydroxy-$ Comp. 30 24-oxo-24-[(2-sulfoethyl)amino]cholan-3-yl]amino]-2-oxoethyl]amino]-2-oxoethyl]-1,4,-7,10-tetraazacyclododecane-1,4,7-triacetic 25 acid; $(3\beta, 5\beta, 7\alpha, 12\alpha) - 3 - [[[[4,7,10-tris(carboxyme-$ Comp. 31 thyl)-1,4,7,10-tetraazacyclododecyl]acetyl]amino]acetyl]amino]-7,12-dihydroxy-cholan-24oic acid; $N^2 - [(3\alpha, 5\beta, 7\alpha, 12\alpha) - 3, 7, 12 - trihydroxy - 24 - oxo-$ Comp. 32 30 $cholan-24-yl]-N^{6}-[[4,7,10-tris(carboxyme-$

5

10

20

30

107

thyl)-1,4,7,10-tetraazacyclododecyl]acetyl]-L-lysine

- Comp. 33 (3β,5β,7α,12α)-3-[[6-[[[4,7,10-tris(carboxy-methyl)-1,4,7,10-tetraazacyclododecyl]ace-tyl]amino]-1-oxohexyl]amino]-7,12-dihydroxy-cholan-24-oic acid;
- Comp. 34 [[(3a,5\beta,7a,12a)-3-[[3-[4,7,10-tris(carboxy-methyl)-1,4,7,10-tetraazacyclododecyl]-2-hy-droxypropyl]oxy]-7,12-dihydroxy-cholan-24-oic acid;
- Comp. 35 [[(3\beta,5\beta,7\alpha,12\alpha)-3-[[5-[4,7,10-tris(carboxy-methyl)-1,4,7,10-tetraazacyclododecyl]-4-hy-droxy-1-oxopentyl]amino]-7,12-dihydroxy-cholan-24-oic acid.
- 15 15. Contrast diagnostic pharmaceutical compositions comprising at least one of the complex chelates according to claims 1 to 14 or a salt thereof.
 - 16. Pharmaceutical composition according to claim 15, to obtain images of organs and/or tissues of human and animal body, through the use of nuclear magnetic resonance.
 - 17. The use of the complex chelates of the compounds of formula (I) or of the salts thereof for the preparation of diagnostic formulations to obtain images
- of organs and/or tissues of human and animal body, through the use of nuclear magnetic resonance.
 - 18. The use of the complex chelates of the compounds of formula (I) or of the salts thereof for the preparation of diagnostic formulations to obtain images of the hepatobiliary system of human and animal body, through the use of nuclear magnetic resonance.

INTERNATIONAL SEARCH REPORT

Intern. In Application No PCT/EP 95/01958

		PCT/EP 9	5/01958
A. CLAS IPC 6	SIFICATION OF SUBJECT MATTER A61K49/00 A61K51/04		
According	to International Patent Classification (IPC) or to both national cla	assification and IPC	
1	OS SEARCHED	anneadon alle II C	
Minimum IPC 6	documentation searched (classification system followed by classifi $A61K$	cation symbols)	
Documenta	ation searched other than minimum documentation to the extent th	at such documents are included in the fields	searched
Electronic o	data base consulted during the international search (name of data	base and, where practical, search terms used)
C. DOCUM	MENTS CONSIDERED TO BE RELEVANT		
Category °	Citation of document, with indication, where appropriate, of the	relevant passages	Relevant to claim No.
X	EP-A-O 417 725 (HOECHST AG.) 20 cited in the application see claims see page 3 - page 11	March 1991	1
X	EP-A-0 279 307 (ABBOTT LABORATOR March 1991 cited in the application see page 4, line 40 - page 7, lisee page 14, line 12 - line 26; see page 12 see page 13, line 48 - line 57	ine 24	1-18
A	US-A-5 169 944 (NELSON JAMES A December 1992 see column 5, line 52 - line 61;	•	1-3, 15-18
X Furth	ner documents are listed in the continuation of box C.	X Patent family members are listed	in annex.
° Special cate	egories of cited documents :		
"A" docume consider diffiling da" "L" documer which is	rent defining the general state of the art which is not ired to be of particular relevance locument but published on or after the international	"T" later document published after the into or priority date and not in conflict will cited to understand the principle or the invention of document of particular relevance; the cannot be considered novel or cannot involve an inventive step when the document of particular relevance; the	th the application but heory underlying the claimed invention be considered to cument is taken alone claimed invention
other m 'P" documer	nt published prior to the international filing date but	cannot be considered to involve an in document is combined with one or in ments, such combination being obvior in the art.	ore other such docu- us to a person skilled
	an the priority date claimed	'&' document member of the same patent	
	September 1995	Date of mailing of the international se	arch report
Name and ma	ailing address of the ISA European Patent Office, P.B. 5818 Patentiaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized officer Berte, M	

INTERNATIONAL SEARCH REPORT

Intern. .nal Application No
PCT/EP 95/01958

(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT					
egory °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.			
	BIOCONJUGATE CHEM. (1991), 2(2), 117-23 CODEN: BCCHES; ISSN: 1043-1802, 1991 BETEBENNER, DAVID A. ET AL 'Hepatobiliary delivery of polyaminopolycarboxylate chelates: synthesis and characterization of a cholic acid conjugate of EDTA and biodistribution and imaging studies with its indium-111 chelate' see page 119; figure 1	1-5,7,8, 10,14,15			
	J. AM. CHEM. SOC. (1989), 111(8), 2900-9 CODEN: JACSAT; ISSN: 0002-7863, 1989 GROVES, JOHN T. ET AL 'Regioselective oxidation catalysis in synthetic phospholipid vesicles. Membrane-spanning steroidal metalloporphyrins' see figures 1,2; table 1	1-5,7,8			
	WO,A,95 19186 (NYCOMED IMAGING A S;MATTHEWS DEREK PETER (GB); KLAVENESS JO (NO);) 20 July 1995 see page 5, paragraph 2 see page 7, paragraph 4 - page 9, paragraph 3; claims	1-18			

1

Form PCT/ISA/210 (continuation of second sheet) (July 1992)

INTERNATIONAL SEARCH REPORT

Information on patent family members

Intern. nal Application No
PCT/EP 95/01958

Patent document cited in search report	Publication date		t family iber(s)	Publication date	
EP-A-417725	20-03-91	DE-A-	3930696	28-03-91	
		AU-B-	637822	10-06-93	
		AU-A-	6244190	21-03-91	
		CA-A-	2025294	15-03-91	
		IL-A-	95668	30-03-95	
		JP-A-	3109396	09-05-91	
EP-A-279307	24-08-88	US-A-	5057302	15-10-91	
		AU-B-	605241	10-01-91	
		AU-B-	1168588	18-08-88	
		DE-D-	3884233	28-10-93	
		DE-T-	3884233	03-03-94	
		ES-T-	2059411	16-11-94	
		JP-A-	63290854	28-11-88	
		US-A- 	5227474	13-07-93	
US-A-5169944	08-12-92	NONE			
WO-A-9519186	20-07-95	NONE			